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Evaluation of Military Field-Water Quality

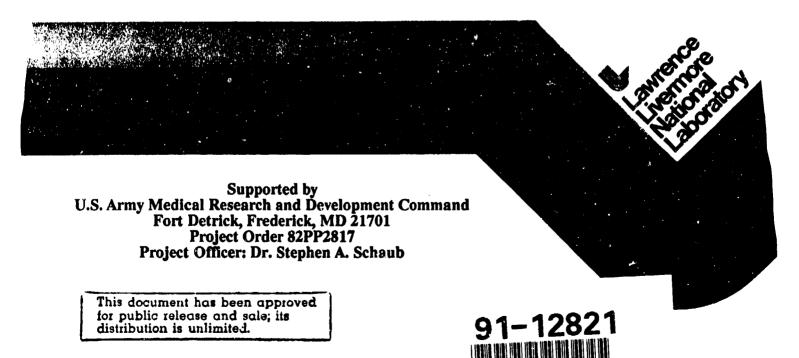
Volume 2. Constituents of Military Concern from Natural and Anthropogenic Sources Part 1. Organic Chemical Contaminants



D. W. Layton
B. J. Mallon
T. E. McKone
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P. C. Lessard

January 1988



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However, we identified several organic solutes that could cause taste and odor problems. These compounds include trichloromethane, ethylbenzene, toluene, tetrachloroethene, and chlorinated phenols. Additionally, geosmin and 2-methylisoborneol, which are two metabolites of blue-green algae (cyanobacteria) and associated gram-negative filamentous bacteria (actinomycetes), can produce taste and odor problems and are of particular concern, especially when algal blooms are present. Oil and grease could also impair the potability of water supplies, and by fouling reverse-osmosis (RO) membranes, oil and grease could impair the performance of reverse osmosis water purification units (ROWPUS).

This report is the first part of the second volume of a nine-volume study entitled Evaluation of Military Field-Water Quality. The second and third parts of this volume address pesticides and inorganic chemicals, respectively. Titles of the other volumes are as follows: Vol. 1, Executive Summary; Vol. 3, Opportunity Poisons; Vol. 4, Health Criteria and Recommendations for Standards; Vol. 5, Infectious Organisms of Military Concern Associated with Consumption: Assessment of Health Risks, and Recommendations for Establishing Related Standards; Vol. 6, Infectious Organisms of Military Concern Associated with Nonconsumptive Exposure: Assessment of Health Risks, and Recommendations for Establishing Related Standards; Vol. 7, Performance Evaluation of the 600-GPH Reverse Osmosis Water Purification Unit (ROWPU): Reverse Osmosis (RO) Components; Vol. 8, Performance of Mobile Water Purification Unit (ROWPU) and Pretreatment Components of the GOO-GPH Reverse Osmosis Water Purification Unit (ROWPU) and Consideration of Reverse Osmosis (RO) Bypass, Potable-Water Disinfection, and Water-Quality Analysis Techniques; and Vol. 9, Data for Assessing Health Risks in Potential Theaters of Operation for U.S. Military Forces.

Evaluation of Military Field-Water Quality

Volume 2. Constituents of Military Concern from Natural and Anthropogenic Sources Part 1. Organic Chemical Contaminants

D. W. Layton, B. J. Mallon, T. E. McKone, Y. E. Ricker, and P. C. Lessard*

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FOREWORD

This report is the first part of the second volume of a nine-volume study entitled Evaluation of Military Field-Water Quality. The second and third parts of this volume address pesticides and inorganic chemicals, respectively. Titles of the other volumes are Vol. 1, Executive Summary; Vol. 3, Opportunity Poisons; Vol. 4, Health as follows: Criteria and Recommendations for Standards; Vol. 5, Infectious Organisms of Military Concern Associated with Consumption: Assessment of Health Risks, Recommendations for Establishing Related Standards; Vol. 6, Infectious Organisms of Military Concern Associated with Nonconsumptive Exposure: Assessment of Health Risks, and Recommendations for Establishing Related Standards; Vol. 7, Performance Evaluation of the 600-GPH Reverse Osmosis Water Purification Unit (ROWPU): Reverse Osmosis (RO) Components; Vol. 8, Performance of Mobile Water Purification Unit (MWPU) and Pretreatment Components of the 600-GPH Reverse Osmosis Water Purification Unit (ROWPU) and Consideration of Reverse Osmosis (RO) Bypass, Potable-Water Disinfection, and Water-Quality Analysis Techniques; and Vol. 9, Data for Assessing Health Risks in Potential Theaters of Operation for U.S. Military Forces.

As indicated by the titles listed above, the nine volumes of this study contain a comprehensive assessment of the chemical, radiological, and biological constituents of field-water supplies that could pose health risks to military personnel as well as a detailed evaluation of the field-water-treatment capability of the U.S. Armed Forces. The scientific expertise for performing the analyses in this study came from the University of California Lawrence Livermore National Laboratory (LLNL) in Livermore, CA; the University of California campuses located in Berkeley (UCB) and Davis (UCD), CA; the University of Illinois campus in Champaign-Urbana, IL; and the consulting firms of IWG Corporation in San Diego, CA, and V. J. Ciccone & Associates (VJCA), Inc., in Woodbridge, VA. Additionally a Department of Defense (DoD) Multiservice Steering Group (MSG), consisting of both military and civilian representatives from the Armed Forces of the United States (Army, Navy, Air Force, and Marines), as well as representatives from the U.S. Department of Defense, and the U.S. Environmental Protection Agency provided guidance, and critical reviews to the researchers. The reports addressing chemical, radiological, and biological constituents of field-water supplies were also reviewed by scientists at Oak Ridge National Laboratory in Oak Ridge, TN, at the request of the U.S. Army. Furthermore, personnel at several research laboratories, military installations, and agencies of the U.S. Army and the other Armed Forces provided technical assistance and information to the researchers on topics related to field water and the U.S. military community.

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The principal investigators at the Lawrence Livermore National Laboratory (LLNL), Drs. Jeffrey Daniels, David Layton, and Lynn Anspaugh, extend their gratitude and appreciation to all of the participants in this study for their cooperation, assistance, contributions and patience, especially to Dr. Stephen A. Schaub, the project officer for this monumental research effort, and to his military and civilian colleagues and staff at the U.S. Army Biomedical Research and Development Laboratory (USABRDL). A special thank you is extended to the editors, secretaries, and administrative personnel of the Environmental Sciences Division at LLNL, particularly to Ms. Barbara Fox, Ms. Yvonne Ricker, Ms. Penny Webster-Scholten, Mr. E.G. Snyder, Ms. Gretchen Gallegos, Ms. Angelina Fountain, Ms. Sherry Kenmille, Mr. David Marcus, Ms. Martha Maser, and Ms. Sheilah Hendrickson, whose efforts, support, and assistance included the typing and editing of over 2500 pages of text.

CONTENTS

Foreword	ii
Acknowledgments	iv
List of Tables	v
List of Figures	vii
Preface	i
Abstract	1
Introduction	2
Overview of the Screening Methodology	2
Identification of Organic Contaminants	8
Screening Concentrations for Toxic and Organoleptic Responses	8
Toxic Effects	8
Organoleptic Responses	7
Screening Based on Measured and Predicted Concentrations of Organic Solutes	7
Measured Solute Concentrations	8
Screening for Systemic Toxicants	g
Screening for Organoleptic Effects	15
Predicted Organic-Solute Concentration Maxima in the Surface Waters of Representative Foreign Countries	23
Application of the Model to Five Representative Countries	28
Biochemical Compounds Released by Algae and Associated Microorganisms	33
Summary and Recommendations	34
References	36
Appendix A: Health Standards and Guidelines for Organic	
Water Contaminants	40
Bibliography	47
Appendix B: Lethal Doses used in the Calculation of Screening Concentrations for Toxicity	48
References	55
Appendix C: Taste and Odor Thresholds	56
References	60
Appendix D: Data on the Concentrations of Organic Solutes in Fresh and Salt Waters	62
References	98
45UAUAU44U4	90

LIST OF TABLES

1.	Summary of the organic solute concentration measurements for 150 organic substances in water supplies located in major geographic regions around the world	9
2.	Ratios of maximum observed concentrations (C_{max}) to screening concentrations (C_s) calculated from oral LD50's for mammalian animal species	11
3.	Comparison of maximum concentration values in the WaterDROP data base with those used in the present study for substances with screening ratios greater than or approximately	
	equal to unity in Table 2	13
4.	Comparison between maximum concentrations and the minimum concentrations for taste- or odor-detection thresholds	16
5.	Comparison between screening concentrations based on LD50 values and all concentration data reported for those organic chemicals in Table 4 having a ratio of maximum concentration to taste- or odor-detection threshold greater than unity	17
6.	Production-output factors for estimating source terms from the 13 industrial/mining categories that we consider to be major sources of organic-chemical pollution in foreign countries	28
7.	U.S. production of organic by-products in g/y for each industrial category	29
8.	Annual average and low dilution volumes provided by major rivers in each of the candidate countries	30
9.	Annual mass discharges of the major organic contaminants released into the raw-effluent wastestreams of all 13 industrial/mining categories in each of five representative countries	30

10.	The estimated maximum and mean concentrations of each of the major organic chemicals in the surface waters of each of five representative countries	32
11.	Toxicity- or organoleptic-threshold concentrations in water $(T_i, based on 70-kg military personnel consuming up to 15 L/d of drinking water) and screening-criteria ratios (S_{i,m}, calculated by dividing the estimated maximum concentration in the water of a particular country, C_{i,m}, by T_i) for each of the major organic chemicals (i) in each of five representative countries (m)$	33
A-1.	Regulatory status of organic compounds (excluding pesticides)	41
B-1.	Oral LD50's for organic water contaminants for most sensitive mammalian species	49
B-2.	Abbreviations used in Table B-1 for exposure mode and species	54
C-1.	Reported taste- and odor-detection threshold concentration for organic chemicals for which we compiled water-quality monitoring data	5 <i>7</i>
D-1.	Fresh-water concentration data for organic substances other than pesticides	63
D-2.	Salt-water concentration data for organic substances other than pesticides	84

LIST OF FIGURES

1.	Diagram of the methodology for identifying organic solutes that could pose health risks to military personnel consuming field waters	4
2.	Possible comparisons between water concentrations and threshold concentrations for toxic and organoleptic responses	4
3.	Log probability plot of the maximum concentration for each organic solute that was reported in one or more studies	10
4.	Influence diagram for processes that affect the concentration of organic solutes in surface waters	25
5.	A schematic view of a national water system for use in screening organic solutes likely to be at potentially toxic levels	26

EVALUATION OF MILITARY FIELD-WATER QUALITY VOLUME 2. CONSTITUENTS OF MILITARY CONCERN FROM NATURAL AND ANTHROPOGENIC SOURCES

Part 1. Organic Chemical Contaminants

PREFACE

Water that may be used by military personnel in the field can contain many different organic and inorganic chemical constituents. These chemicals may exist in a dissolved or colloidal state or on suspended material, and they are present as a consequence of either natural geochemical and hydrological processes or the industrial, domestic, or agricultural activities of man.

The health risk to military personnel from a chemical constituent of field water is largely a function of the frequency with which it occurs at concentrations that are high enough to produce a toxic or organoleptic (e.g., detectable taste or odor) effect that leads directly or indirectly to the diminished ability of exposed military forces to perform assigned tasks. To minimize performance-related effects in military personnel using field-water supplies, the high-risk chemical constituents must be identified and analyzed. The potential health risks of the contaminants can then be managed by adopting and meeting field-water quality standards. The health effects that could occur when standards are exceeded can be addressed on a case-by-case basis.

The objective of Volume 2 of Evaluation of Military Field-Water Quality is to indicate the chemical constituents of field water that are of possible military concern and to describe the screening methodology and supporting data that we used to identify them. Briefly, the screening methodology is separated into two phases. In both phases the general approach consists of comparing (1) the maximum likely concentration in field water of each possible chemical constituent with (2) a corresponding concentration we estimate to be the threshold above which toxic effects, including impaired performance, could occur. Our analyses are based on 70-kg military personnel consuming field water at a maximum rate of 15 L/d. Maximum likely concentrations in field water for each chemical are derived from our compilation of available U.S. and worldwide water-quality monitoring data. However, in the first phase of screening we make conservative assumptions to extrapolate the threshold concentration above which toxic effects could occur in military forces from either oral-mammalian LD50 (lethal dose to 50% of a population) data or Acceptable Daily Intake (ADI) values for humans. The result of this screening procedure is to exclude from further consideration those chemical constituents

that are not expected to be of military concern. Although the conservative assumptions incorporated into the initial screening exercise minimize the omission of substances that may actually be of concern, some substances may be identified incorrectly as high-risk. Therefore, to refine the results of the initial screening effort, we reexamine the available monitoring data and review the published human-toxicity data more carefully for each chemical indicated to be of possible military concern. Next, we use any more appropriate human-toxicity data (e.g., dose-response information from reported accidental poisonings, occupational exposures, or therapeutic administrations) that we find and apply it in the second phase of screening. Then, as in the initial screening procedure, any ratio greater than unity between the maximum likely concentration for a chemical in field water and the concentration above which it could produce toxic or organoleptic effects in 70-kg military personnel consuming field water at a maximum rate of 15 L/d indicates that the chemical really could be of military concern. Because impaired performance can occur as a result of indirect health effects, especially from heat illnesses caused by dehydration resulting from reduced consumption of poor-tasting water, we also screen the initial list of chemicals by comparing maximum likely concentration data for each one with available data corresponding to the concentration of the substance that represents the taste- or odor-detection threshold in water.

To facilitate data acquisition, analysis, and review, as well as application of the screening methodology, we separated the potential chemical constituents of field water into three categories and divided Volume 2 into three corresponding parts. Part 1 covers organic solutes (except pesticides), Part 2 addresses pesticides, and Part 3 focuses on inorganic chemicals.

ABSTRACT

In this part of Volume 2 we focus on the identification and analysis of organic water contaminants that could degrade the performance of military personnel that drink field water. To identify the contaminants of concern, we developed a screening methodology for comparing measured concentrations of organic solutes in U.S. and foreign surface and ground waters against estimated threshold concentrations for toxic and organoleptic effects. If the measured concentration of a substance in water was higher than the concentration used as an effects threshold for toxic or organoleptic responses, we then closely examined the substance's occurrence and toxicity to decide on recommending development of a water standard to protect military personnel from adverse health effects. After comparing the measured concentrations with the threshold concentrations for toxic or organoleptic effects, we concluded that there was only a small probability that troops would experience performance-degrading effects as a result of drinking field water containing organic solutes. However, we identified several organic solutes that could cause taste and odor problems. These compounds include trichloromethane, ethylbenzene, toluene, tetrachloroethene, and chlorinated phenols. Additionally, geosmin and 2-methylisoborneol, which are two metabolites of blue-green algae (cyanobacteria) and associated gram-negative filamentous bacteria (actinomycetes), can produce taste and odor problems and are of particular concern, especially when algal blooms are present. Oil and grease could also impair the potability of water supplies, and by fouling reverse-osmosis (RO) membranes, oil and grease could impair the performance of reverse osmosis water purification units (ROWPUs).

INTRODUCTION

The objective of Volume 2 of Evaluation of Military Field-Water Quality is to identify those water contaminants that have potential for impairing troop performance. Key considerations in identifying hazardous contaminants are their occurrence in foreign water supplies, the concentrations measured, and importantly, their toxicity. Volume 2 is divided into three parts: Part 1 covers organic solutes (except pesticides); Part 2 deals exclusively with pesticides; and Part 3 addresses inorganic solutes.

Of particular concern are health effects that would directly affect an individual's ability to conduct a military mission. Indirect effects, such as heat illnesses caused by dehydration resulting from the reduced consumption of poor-tasting water, are also important. The health risk of an organic contaminant in drinking water is a function of its occurrence, expected concentrations, toxicity, and organoleptic properties. As an illustration, a substance that is toxic at low concentrations in water but is found rarely in surface and ground waters would not be considered a high-risk substance. Accordingly, a basic requirement of a screening methodology for field-water contaminants is that it include those factors that directly influence risk. For example, this methodology must consider occurrence in different water sources, measured or predicted concentrations in water, and finally, concentrations that could cause adverse health responses. Another requirement is that the screening procedure should minimize omission of organic solutes that are actually of concern. This means that the methodology should be based on conservative assumptions, even if some substances are incorrectly identified as high-risk during the initial phases of screening. In the following sections we review the basic methodology for screening organic solutes, the data and procedures used to represent contaminant concentrations, and the data and procedures used to estimate threshold-effect concentrations.

OVERVIEW OF THE SCREENING METHODOLOGY

The basic procedure for screening an organic solute, as shown schematically in Fig. 1, is to compare its measured or predicted concentration in water with a screening concentration which represents a no-effect level; that is, the screening concentration has a low probability of degrading performance or causing an organoleptic response. Figure 2 depicts the various comparisons that can result. If the solute concentration is below the screening concentration for toxic effects and the organoleptic concentration (comparison A), then the substance does not constitute a <u>potential</u> health risk. However, if the concentration is above either of these screening concentrations (i.e., comparisons B to E),

then the substance is considered a potential high-risk contaminant. The highest potential health risk occurs when the odor or taste threshold is above the no-effect threshold, because there is no organoleptic warning of possible danger (comparison E). The second phase of the screening methodology is a more detailed analysis of the toxicity and occurrence of these high-risk solutes to ensure that they are identified correctly.

The presence of organic solutes in natural waters is a complex function of usage, pollution controls, environmental chemistry (e.g., solubility, volatility, decomposition rates, etc.), and transport (i.e., dilution and diffusion) in surface and ground waters. Measured concentrations of a solute reflect all of the above factors. If we knew all of the relevant parameters for different organic compounds, as well as site-specific hydrologic characteristics, we could conceivably predict concentrations in various water supplies. Unfortunately these data are incomplete for most chemicals. Therefore, in order to estimate the probable occurrence of organic chemicals in field-water supplies, we have relied principally on measured concentrations in treated and untreated waters in foreign countries and the U.S. To augment the measured concentration values, we have used a methodology to predict levels of organic contaminants in the surface waters of foreign countries receiving industrial waste waters.

One of the more difficult components of the screening methodology is the calculation of threshold solute concentrations, above which there would be a high probability of toxic effects in humans. Acquisition and analysis of toxicological data on each substance to define these thresholds is an expensive proposition. An alternative is to estimate the thresholds from the more widely available data on mammalian species. The most frequently reported parameter in this regard is the lethal dose (expressed in mg of chemical per kg of body weight) to 50% of a population of laboratory animals (i.e., an LD50). To estimate human threshold concentrations from mammalian data, we use the LD50 of the species most sensitive to the substance of concern. This value is multiplied by a fraction derived from a statistical analysis of the ratios of no-effect dose rates [in mg/(kg•d)] to LD50's for a set of chemicals. After the applicable dose is calculated, an equivalent concentration in water is calculated, based on a 70-kg man consuming 15 L of water per day. The calculated threshold concentrations are then compared with observed or predicted concentrations in water (see Fig. 1). Those substances that meet the high-risk screening criteria are then scrutinized further to determine whether they are indeed compounds that pose adverse health risks to soldiers that drink field water.

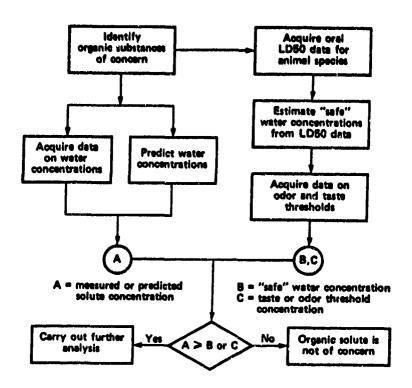


Figure 1. Diagram of the methodology for identifying organic solutes that could pose health risks to military personnel consuming field waters. The basic procedure for screening a chemical is to compare measured or predicted concentrations in water with concentrations that represent a no-effect level. Solutes whose concentrations in water are above toxic or organoleptic thresholds are potentially high-risk compounds.

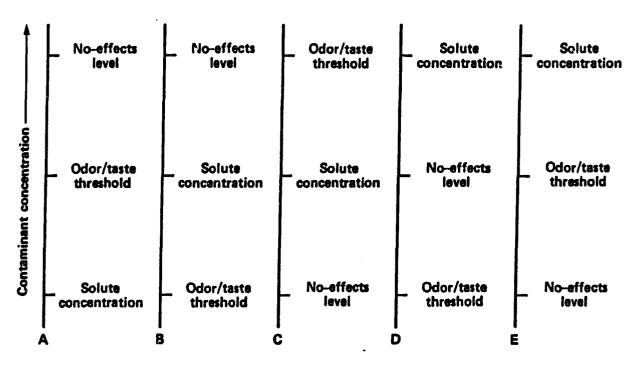


Figure 2. Possible comparisons between water concentrations and threshold concentrations for toxic and organoleptic responses.

IDENTIFICATION OF ORGANIC CONTAMINANTS

It would be impossible to complete screening analyses on all organic compounds that are potential water contaminants. Accordingly, the goal of our data-gathering efforts was to focus on those substances that might actually constitute a health threat to troops drinking treated or untreated field water. The substances identified for screening analyses included organic substances for which health standards had been established by domestic or foreign governments or scientific organizations and substances that appeared on hazardous material lists. Our premise was that such standards and lists reflect objective and subjective evaluations of the health risks of various organic solutes in water supplies worldwide. We supplemented this set of compounds with substances that were reported in foreign water supplies and substances that were measured frequently in U.S. water supplies. The primary emphasis, though, was on measured concentrations of organic solutes in foreign water resources because these supplies are the most likely sources of field drinking waters. The final screening list consisted of over 200 potentially hazardous substances (see Appendix A).

SCREENING CONCENTRATIONS FOR TOXIC AND ORGANOLEPTIC RESPONSES

In this section we describe the methodology we used to derive the minimum concentrations of organic solutes in field water that could produce toxic effects in military personnel consuming up to 15 L of water per day. These minimum concentrations represent the screening concentrations for toxic effects. We also discuss the data used to identify contaminants that could cause organoleptic responses in exposed individuals.

TOXIC EFFECTS

A major challenge in developing a methodology for identifying organic solutes that could pose adverse health risks to soldiers is the quantification of a screening concentration for toxic effects. Ideally, we would calculate the equivalent screening concentration for toxicity as follows:

$$C_s = \frac{NOEL}{SF} \cdot \frac{W}{I}$$
, where (1)

C_s = screening concentration for human toxicity, mg/L; NOEL = maximum no-observed-effect level in rats, mg/(kg·d); SF = safety factor for adjusting the animal data to compensate for intra- and interspecies variation, dimensionless;

W - reference body weight of a soldier, kg; and

I - maximum daily intake of water, L/d.

Unfortunately, the no-observed-effect levels (NOEL's) derived from animal studies are available for only a relatively small number of substances. The data that best meet the criteria for availability are the oral LD50 data for mammalian species. The LD50, however, is not particularly useful for developing a screening dose or concentration because there is no direct way to relate this acute measure of animal toxicity (i.e., lethality) to an effects threshold in humans. To relate LD50 data to screening concentration thresholds, we adopted a statistical approach in which an LD50 is used to define a lower-bound limit of toxic effects, based on a statistical analysis of the relationship between LD50 values and subchronic (90-d) NOEL's. McNamara. 1 in a study of the relationship between chronic and acute toxicities, computed the ratios of subchronic NOEL's [mg/(kg·d)] to LD50's (oral administrations to rats, mg/kg) for 33 substances. The geometric mean of the log normally distributed ratios was 0.03 d⁻¹ with a geometric standard deviation of 4.8. A conservative NOEL (i.e., a NOEL that has a high probability of being below a toxic threshold) can then be calculated from an LD50 by multiplying the median lethal dose by a ratio that corresponds to a suitable cumulative percentile on the lognormal distribution. For the purposes of this screening exercise, we chose the ratio corresponding to the 10th cumulative percentile of the lognormal distribution, or $0.004 \, d^{-1}$ (i.e., $0.03/4.8^{1.3}$).

The 90-d exposure period used for the NOEL is similar to a human exposure period of nearly ten years, based on the assumption that the ratios of the two periods to the lifespans of the respective species are approximately the same (i.e., 90 d/730 d = 9 y/75 y). By comparison, the assumed exposure period for consuming field-water supplies is one year, and so there is an implicit margin of safety in using the derived subchronic NOEL. A safety factor of 100 was selected to adjust the NOEL because this factor has been widely used by regulators for standards setting, and it is supported by data on inter- and intraspecies variations (see Dourson and Stara²). The reference body weight of a soldier is 70 kg, and the maximum intake of drinking water is 15 L/d. In addition, we assume that the military population is predominately male; age ranges from 18 to 50 y; and the troops are in good health. We use these assumptions to calculate the screening concentration, in mg/L, from an animal LD50 (mg/kg) as:

$$C_{s} = \frac{LD50 \cdot 0.004}{100} \cdot \frac{70}{15} . \tag{2}$$

To simplify the computation of the screening concentrations, we multiplied the LD50 values by 10⁻⁴ to obtain a C_s value. In addition, we used the oral LD50 value of the most sensitive mammalian species; most of these LD50's were for rats. When oral LD50's (or intraperitoneal LD50's in some instances) were not listed in the <u>Registry of Toxic Effects of Chemical Substances</u>, we used the lowest lethal doses for oral administrations. Appendix B contains the animal LD50 data used to calculate screening concentrations.

ORGANOLEPTIC RESPONSES

Organic substances dissolved in water can impart undesirable tastes and odors. However, data on the organoleptic properties of organic solutes rarely deal with the objectionability of the tastes and odors. Instead, data are presented on the concentrations at which the odor or taste of a chemical is detected by an individual or panel of testers. Threshold concentrations are reported as individual values, ranges, means, medians, or minima, making it almost impossible to accurately compare the organoleptic properties of chemicals. Moreover, because researchers have not used the same techniques to measure organoleptic responses, the resulting odor and taste thresholds can show large variability -- even for the same compound. 4 With these difficulties in mind, our approach is to screen the organic chemicals that are potential contaminants of field water, based on their organoleptic properties, by comparing the lowest odor or taste threshold recorded in the literature with the measured or predicted solute concentrations for the chemicals in natural waters (see Appendix C for a listing of the organoleptic data). Substances whose concentration in water is above the measured odor or taste threshold are identified as potential high-risk compounds from an aesthetic standpoint. Once identified, these compounds must be analyzed further to determine whether their tastes or odors are likely to cause adverse responses. In such cases we reviewed the applicable literature on the particular compound.

SCREENING BASED ON MEASURED AND PREDICTED CONCENTRATIONS OF ORGANIC SOLUTES

The screening methodology employs a comparison between our estimated toxic/organoleptic concentration and predicted and measured concentrations of organic substances in water resources located in various parts of the world. Most of our effort was directed toward the acquisition of data on organic solutes in foreign water supplies. Unfortunately, such data were difficult to locate and obtain. One reason for the absence

of data on organic solutes is the lack of the appropriate analytical equipment and water-sampling programs in developing countries, and even industrialized ones. Because of this, we were forced to supplement the foreign data with U.S. data. Whether U.S. data are reasonable surrogates for foreign water supplies is debatable. Nevertheless, the large number of organic substances entering U.S. surface and ground waters from industrial and domestic sources would seem to bracket the kinds of compounds that might be encountered elsewhere; however, less-developed countries are likely to have different industrial mixes (and therefore different waste effluents) and less stringent water pollution controls, which would result in higher solute concentrations. To ensure that the monitoring data we were able to obtain provided a reasonable picture of water contamination expected in foreign water supplies, we also calculated concentrations of organic solutes in surface waters of five countries that receive waste waters from a mix of different industries.

MEASURED SOLUTE CONCENTRATIONS

In order to characterize the concentrations of organic solutes that are apt to occur in foreign field-water supplies, we collected concentration data for many different countries. Table 1 summarizes the measured concentrations for 150 organic substances in major geographic regions of the world. Most of our efforts were directed toward the acquisition of water-quality data for water supplies outside of North America, but we have also included a significant number of samples from the U.S. and Canada to supplement the foreign data. We encoded the appropriate water-quality data for entry into our computerized data base. After the entries were made, we checked the values in the data base against the original values in the source documents. We also analyzed the concentration data statistically to identify unusually high or unusually low values. Whenever such concentrations were identified, we examined the source document to determine their validity. If the validity of the reported concentrations could not be established, we removed the values from the data base. Appendix D contains the concentration data that we were able to obtain from the open literature. Figure 3 is a log probability plot of the maximum concentrations for each of the solutes measured in different studies. This plot clearly shows that nearly all organic solute concentrations are below 1 mg/L.

Table 1. Summary of the organic solute concentration measurements for 150 organic substances in water supplies located in major geographic regions around the world.

Geographic location	Num	ber of concentration values selected for screening
p e		364
North America		235
Southeast Asia		96
Africa		77 .
Middle East		35
Asia		34
South America		20
Central America		8
	Total	869

Screening for Systemic Toxicants

The objective of the first phase of the screening analysis was to identify compounds that could pose health risks by virtue of their frequency of occurrence, measured concentrations, and toxicity. In Table 2 we present the ratios of the maximum measured concentration of each organic solute to the screening concentration we estimated as the threshold for toxicity. The compounds of greatest concern are those with ratios greater than unity (to be conservative, we also examined substances with ratios greater than 0.1). To ensure that the maximum concentration values for compounds with a screening ratio greater than 1 were valid, we compared these values with those independently recorded for the same chemicals in a separate data base named WaterDROP. This data base was prepared under the sponsorship of the U.S. Environmental Protection Agency (EPA). It contains data on over 20,000 occurrences of nearly 1000 organic solutes in different kinds of waters (e.g., industrial effluents, process streams, surface and ground waters. etc.).⁵ We used WaterDROP only as an auxiliary check of our concentration data because we were unable to verify the accuracy of the reported data as waterDROP is no longer supported as a sponsored project of EPA. In Table 3 we show those compounds that had maximum concentrations in WaterDROP greater than the ones we reported and also those compounds with values less than our maximum values.

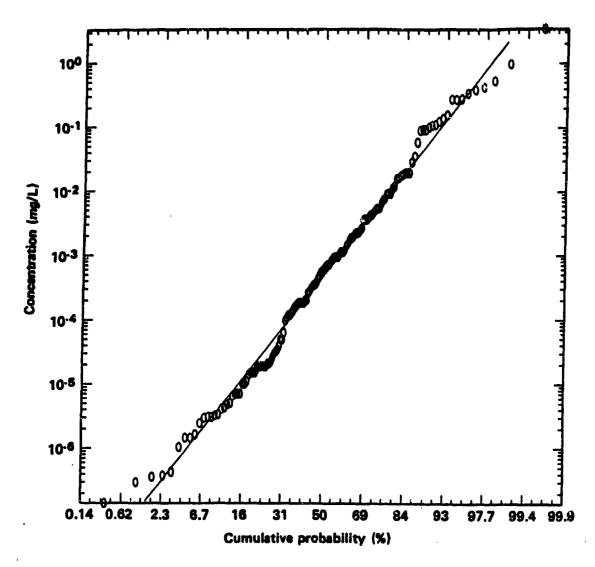


Figure 3. Log probability plot of the maximum concentration for each organic solute that was reported in one or more studies. The geometric mean concentration was 0.0005 mg/L, with a geometric standard deviation of 40. The total number of data points (n) equals 150.

The high concentrations of compounds in surface waters reported in WaterDROP were all in conjunction with immediate (e.g., adjacent to discharge points) industrial pollution. For example, phenol and xylene were both found in surface waters directly receiving effluent from a petrochemical plant. Dilution would diminish these concentrations.

The compound with the highest ratio is trisodium nitrilotriacetate (NTA). Data on the concentrations of NTA were obtained from a study by Woodiwiss et al., 6 who sampled several Canadian streams receiving sewage effluents from cities in the province of Ontario. The NTA was present in detergents as a replacement for phosphates. The

Table 2. Ratios of maximum observed concentrations (C_{max}) to screening concentrations (C_s) calculated from oral LD50's for mammalian animal species. The reference code is an abbreviation for the reference containing the concentration data (see Appendix D).

Substance	Water type ^a	C _{max} (mg/L)	C _s b (mg/L)	Ratio C _{max} /C _s	Ref.
Phenol, m-chloro-	sw	6.00E-03	5.70E-02	1.05E-01	WEGM79
Phthalate, butylbenzyl	GW	3.80E-02	3.16E-01 ^C	1.20E-01	EPA811
Phenol, 4-chloro-	SW	3.90E-03	2.61E-02	1.49E-01	WEGM79
Aniline, 2-chloro-	sw	3.90E-03	2.56E-02	1.52E-01	WEGM81
Ethane, trichlorotrifluoro-	GW	1.35E-01	8.60E-01 ^C	1.57E-01	EPA811
Ethene, chloro- (vinyl chloride)	FDW	8.40E-03	5.00E-02	1.68E-01	WEST84
Ether, bis(2-chloroisopropyl)	SW	4.60E-03	2.40E-02	1.92E-01	MEIJ76
Benzene, 1,2-dichloro-	sw	1.00E-02	5.00E-02	2.00E-01	PIET80
Ethane, 1,2-dichloro-	FDW	9.80E-03	4.89E-02	2.00E-01	WEST84
Propane, 1,2-dichloro-	FDW	2.10E-02	8.60E-02	2.44E-01	WEST94
Cyclopentadiene, hexachloro-	SW	2.90E-03	1.13E-0?	2.57E-01	COTR83
Benzene	GW	1.00E-01	3.80E-01	2.63E-01	DUIJ81
Benzene, 1,4-dichloro-	GW	1.76E-02	5.00E-02	3.52E-01	PAGE82
Phthalate, di-n-butyl	GW	4.70E-01	1.20E+00	3.92E-01	EPA81
Ethene, tetrachloro-	GW	3.75E-01	8.10E-01	4.63E-01	EPA81
Toluene	GW	3.00E-01	5.00E-01	6.00E-01	DUIJ81
Aniline	SW	1.20E-02	1.95E-02	6.15E-01	WEGM81
Methane, dibromochloro-	FDW	6.30E-02	8.00E-02	7.88E-01	WEST84
Benzene, ethyl-	GW	3.00E-01	3.50E-01	8.57E-91	DUIJ81
Phenol	SW	7.90E-03	8.00E-03 ^C	9.88E-01	SCHO81
Methane, dichloro-	SW	1.70E-02	1.67E-02	1.02E+00	COTR83
Methane, tribromo-	SW	1.17E-01	1.15E-01	1.02E+00	PELE81
Xylene	GW	6.00E-01	4.30E-01	1.40E+00	DUIJ81
Ethene, 1,2-dichloro-	FDW	1.20E-01	7.76E-02	1.56E+00	WEST84
Phenol, pentachloro-	SW	1.00E-02	5.00E-03	2.00E+00	WEGM79
Ethene, trichloro-	GW	1,10E+00	4.92E-01	2.24E+00	ZOET78
Methane, bromodichloro-	FDW	1.10E-01	4.50E-02	2.44E+00	WEST84

Table 2. (Continued)

Substance	Water type ^a	C _{max} (mg/L)	(mg/L)	Ratio C _{max} /C _s	Ref.
Carbon disulfide	SW	3.90E-03	1.40E-03	2.79E+00	KAI83
Ethane, 1,1,1-trichloro-	GW	3.10E-01	7.50E-02	4.13E+00	EPA81
Ethene, 1,1,-dichloro-	GW	9.62E-02	2.00E-02	4.81E+00	PAGE82
Methane, trichloro-	FDW	4.30E-01	8.00E-02	5.38E+00	WEST84
Trisodium nitrilotriacetate	SW	3.36E+00	8.81E-02	4.93E+01	WOOD79

⁸ SW = surface water, GW = ground water, FDW = finished (i.e., treated) drinking water.

concentration producing the ratio greater than 1 was obtained by sampling receiving waters below a sewage outfall. We do not believe that this substance poses a high risk for troops for the following reasons: first, the maximum concentration measured was considerably higher than the other values reported in the study (see Appendix D); second, we did not find any concentration data reported for other countries, suggesting that this material is probably not widely used; and finally, the available toxicity data suggest that it is not particularly toxic. Nixon⁷ completed a toxicity evaluation of NTA and found that a no-observed-effect level during a 90-d exposure to rats was about 200 mg/(kg·d). If we assume an animal-to-man extrapolation (safety) factor of 100 and apply Eq. l, the equivalent drinking-water concentration representing a toxicity threshold for a 70-kg soldier consuming 15 L/d is

$$\frac{200 \text{ mg}}{\text{kg} + \text{d}} \cdot \frac{70 \text{ kg}}{1} \cdot \frac{\text{d}}{15 \text{ L}} \cdot \frac{1}{100} = 9 \text{ mg/L}. \tag{3}$$

Because this value is higher than the maximum measured concentration, there is a strong possibility that even in worst-case exposure conditions, troops would not be affected.

It is also difficult to make a strong case that any of the other compounds with ratios greater than 1 pose a significant health risk to troops. For example, trichloromethane (i.e., chloroform), the compound with the second highest screening ratio, is found in many drinking waters as a product of chlorination, but it rarely reaches concentrations as high as 0.4 mg/L. Moreover, the human toxicity data indicate that dose rates of well

b The screening concentration is equal to the LD50 multiplied by 0.0001.

^C If oral LD50's were not available, then an LD50 resulting from intraperitoneal administration was used; and, if neither oral nor intraperitoneal LD50's were available, the oral LDL0 was used.

Table 3. Comparison of maximum concentration values in the WaterDROP data base with those used in the present study for substances with screening ratios greater than or approximately equal to unity in Table 2.

•	W	aterDROP a		Our Surv	ey
Compound	Water type ^b	Maximum concentration (mg/L)		Maximum concentration (mg/L)	Water type ^b
Phenol	SW	60	>	0.008	SW
Methane, dichloro-	SW	0.03	>	0.017	SW
Phenol, pentachloro-	SW	0.4	>	0.01	SW
Kylene	SW	8.0	>	0.60	GW
Ethene, trichloro-	SW	0.1	<	1.1	GW
Ethane, 1,1,1-trichloro-	SW	0.003	<	0.3	GW
Ethene, 1,1-dichloro-	sw	0.004 ^a	<	0.096	GW
Ethene, 1,2-dichloro-	SW	0.003	<	0.12	FDW
lethane, tribromo-	SW	0.002	<	0.12	SW
Methane, bromodichloro-	FDW	0.021	<	0.11	FDW
Methane, trichloro-	FDW	0.152	<	0.43	FDW

a WaterDROP does not list concentration data for carbon disulfide and trisodium nitrilotriacetate, nor does WaterDROP contain concentration data for Ethene, 1,1-dichloro- specifically (i.e., reported concentration is for unspecified species of dichloroethylene).

over 1 mg/(kg·d) do not produce debilitating effects. The equivalent drinking-water concentration for toxicity for the reference exposure case (i.e., a 70-kg person consuming 15 L/d) is about 5 mg/L, well above the maximum concentration of 0.4 mg/L, and therefore it is unlikely that this substance presents an unacceptable human-health risk.

As a way of checking our screening concentrations for the chlorinated methanes, ethanes, and ethenes, we compared them with toxic concentrations for water calculated from air-concentration threshold limit values (TLV's) for each of these chemicals. A TLV is an air concentration standard that is set to protect worker health for 8-h exposures to toxic substances during a 5-d work week. We converted the TLV's to an equivalent water concentration by adjusting for continuous exposures (i.e., a 7-d, 24-h exposure instead of a 5-d, 8-h exposure) and by using an inhalation volume of 10 m³ for a 70-kg man working 8 h. We also assumed that the substances were completely absorbed after inhalation

b FDW - finished drinking water, OW - ocean water, SW - surface water, GW - ground water.

and ingestion. As an example, the TLV of chloroform is 10 parts per million by volume (ppmv or 10 $\text{m}^3/10^6 \text{ m}^3$), and it is equivalent to a concentration of 25.8 mg/L in water for our reference exposure case.

Conc =
$$\frac{10 \text{ m}^3}{10^6 \text{ m}^3} \cdot \frac{10 \text{ m}^3}{\text{d}} \cdot \frac{\text{mole}}{0.022 \text{ m}^3} \cdot \frac{119.4 \text{ g}}{\text{mole}} \cdot \frac{1000 \text{ mg}}{\text{g}} \cdot \frac{5 \text{ d}}{\text{wk}} \cdot \frac{\text{wk}}{7 \text{ d}} \cdot \frac{\text{d}}{15 \text{ L}} = 25.8 \frac{\text{mg}}{\text{L}}$$

This value is more than a factor of five higher than the concentration we derived from oral exposures (i.e., 5 mg/L). The TLV and equivalent concentration in water for 1,1-dichloroethene are 5 ppmv and 10.5 mg/L, respectively. By comparison, the screening concentration calculated from LD50 data is 0.02 mg/L (see Table 2). In a similar fashion, the TLV and equivalent concentration in water for 1,1,1-trichloroethane, trichloroethene, 1,2-dichloroethene, and dichloromethane, are 350 ppmv and 1008 mg/L, 50 ppmv and 142 mg/L, 200 ppmv and 419 mg/L, and 100 ppmv and 184 mg/L, respectively. All of these equivalent concentrations are far above the screening concentrations we used. Our evaluation of the other compounds with screening ratios greater than 1 indicates that the screening concentrations are reasonable for those compounds as well.

We also used acceptable daily intake (ADI) values to check our screening concentrations. The ADI for pentachlorophenol, for example, was estimated in an EPA document to be approximately 2 mg, which translates to a drinking-water concentration of 0.1 mg/L--a factor of 20 higher than the value we used (see Table 2). The ADIs for toluene and phenol (assuming a 70-kg man) have been estimated to be 29.5 and 7 mg, respectively. The equivalent drinking-water concentrations are 2 and 0.5 mg/L. The value for toluene is only a factor of 4 greater than the estimated screening concentration we used; however, the maximum concentration was only 0.3 mg/L.

Data on the toxicities of the bromomethanes (i.e., dibromochloromethane, tribromomethane, and bromodichloromethane) are meager, and it is not possible to derive equivalent concentrations in water for the purpose of verifying our screening concentrations. (See the National Research Council report¹² for brief toxicity reviews of these substances.) However, we note that the bromomethanes have low water solubilities, and therefore it is conceivable that they will not reach concentrations that would be harmful.

The remaining substances with screening ratios greater than 1 (i.e., aniline, ethylbenzene, xylene, and carbon disulfide) have screening concentrations that are below concentrations calculated from the TLV's of those substances. Moreover, the screening concentrations in Table 2 for aniline (0.02 mg/L), ethylbenzene (0.35 mg/L), and xylene (0.43 mg/L) are close to the standards appearing in Table A-1 of Appendix A, that is, 0.1, 1.4, and 2 mg/L, respectively.

Screening for Organoleptic Effects

Organic compounds that impart tastes and odors to field drinking-water supplies are of concern for two basic reasons: troops are apt to become dehydrated as they reduce consumption of the aesthetically poor water, or they may seek out and consume unauthorized water supplies that are more potable, but contain toxic substances or disease-causing organisms. The process of identifying compounds that could induce these types of behavior is difficult because most of the data on odor and/or taste thresholds are based on detection thresholds rather than behavioral responses like "I could not drink this water" (see Fig. 2, Columns D and E). For screening purposes, we assume that the taste/odor thresholds are indicative of a potentially negative response. In addition, we used the minimum threshold concentration measured in water for comparisons (threshold values calculated from air data were used when threshold concentrations in water were unavailable). In Table 4 we present the screening ratios of the maximum concentrations in water (from Appendix D) to taste or odor thresholds in water (from Appendix C). Eighteen compounds had ratios greater than 1, and in Table 5 we list all of the concentration data for those compounds, arranged in alphabetical order.

From these two tables the following information about the observed concentrations can be drawn:

Benzene, 1,4-dichloro-

Only two studies reported concentrations above or equal to the organoleptic limit. The highest concentration was from a survey of New Jersey wells.

Benzene, ethyl-

Occurrence is worldwide, but the only concentration above the taste/odor threshold in Table 5 was for a landfill leachate in the Netherlands.

Biphenyl

The only concentration above the organoleptic threshold was associated with an industrial-sewage outfall into a Norwegian river. The reported concentration near this outfall was two orders of magnitude higher than the next highest reported value (see Table 5).

Carbon disulfide

Of the two concentrations in the data base, only one was above the screening concentration.

Table 4. Comparison between maximum concentrations and the minimum concentrations for taste- or odor-detection thresholds.

CASID	SUBSTANCE	MAXIMUM CONCENTRATION	TASTE/ODOR TYPE	MAXCONC/TASTEDDOR
783334571403173524628424124424867525342475453652677570566992430062295775858754677584381994675844488070757575878886413577778758955278375878786688494658844481199111551158780778780778787878	SUBSTANCE CYCLOMEXANE PHENORITHE AND THE MENORITH AND THE MENORITHMENORIT	Non-19-1-18-14 NEW-9-1-18-19-19-19-19-19-19-19-19-19-19-19-19-19-	CC	784444443333333333333333333333333333333

O = ODOR T = TASTE M = MINIMUM A = AVERAGE H = HATER

Table 5. Comparison between screening concentrations based on LD50 values and all concentration data reported for those organic chemicals that appear in Table 4 and have a maximum concentration to taste- or odor-detection threshold ratio greater than unity.

CASID		SUBSTANCE	TYPE*	EPORTED	SCREENI	* REFERENCE	PLACE
1	6 6 9 1			CONCENTRAT	IONS MG/L	1 .	
106467	BENZEHE,	1,4-DICHLORG	NONNA NONNA			3 SCEETS 2 SCEETS 3 SCEETS 3 SCEETS 3 SCEETS	SHITZERLAND SHITZERLAND SHITZERLAND SHITZERLAND SHITZERLAND
			NTTE NATE			PIESTS HESTS HESTS HESTS	HETHERLANDS UNITED STATES UNITED STATES UNITED STATES
100414	BENZENE,	ETHYL-	3556	20E-0-000	3.50E+00	PAGES SAUE7	HEXIC HEXIC
			555	00-1-0-1-0-1-0-1-0-1-0-1-0-1-0-1-0-1-0-		SAUE	GULF OF MEXICO
			553			SAUE7 SAUE7 SAUE7	MEXIC
			558	130E-0		SAUES	EXIC
			888 888			SACEST 1 SACEST 1 TRUSSE	MEXIC MEXIC
			3 57	. 500 . 540 . 500 . 500 . 500		TRUSS	AUSTRALIA GULF OF HEXICO
			ጅጅ	000		TRUS8	T Z
			ZZ:	. 50m-0		321 MEST84	UNITED STATES
			:			SU MESIS 1 VANDS FPAS1	NDS
			38	000-00		1 VANDS1	25
4 7676	SIFRENT				2.40E+00	BENG79	CAMADA, EASTERN ONTARIO
			ZZ			86:079 86:079	CANADA, EASTERN
						BEHO79 BEHO79	CAHADA, EASTERN CAHADA
					•	BERIO79 BERIO79 BERIO79	CAHADA, EASTERN CAHADA, FASTERN
				. 20E-0		1 KVES82 BEN079 1 KVES82	

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SID	CASID	TYPE#	1 6		* REFERENCE	PLACE
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5150	ı W	E S	800 000 000	1.40E-02	1 KVES8 1 SCHOS 7 KAISS	FINLAND HORSAY NEW YORK
	CYCLOPENTADIENE, HEXACHLORG-	XX:	000 000	13E-0	KAISB	.0.
	EINEME, IEINAUMLONU-	i Subs	900 Mmr	. 1 OE+0	S SCHITS S SCHITS SAUERI	SHITZERLAND SHITZERLAND GULF OF MEXICA
		r. V	200 000 000 000 000		SCHH79	AZ C
					1 SCH479	544
		, K.			17 KAIS83 1 CORR77 1 CORR77	HEH YORK NETHERLANDS DENMARK
		in.			TRUSBO TRUSBO VAREBAS	ě
		ine.			1 CORRT7	
		KIX:			VARKSON VARKSON	SOUTH AFRICA SOUTH AFRICA SOUTH AFRICA
		in.	نفن		VANR82	X0
		N. N.	1.00 0.00 0.00 0.00 0.00 0.00		VAHRBZ	zaz Zaz
	•	KKI	2004 2004 2006		1 CORR77	DEHMARK SOUTH AFRICA METHEDI ANDS
		Z.			1 CORR77	
		ZZI VVV			CORR77	METHERLANDS Denhark Serie of
		35.			CORRYY CORRYY CORRYY	DEIXARK
		Ser.			PEAR75	ENGLAND, THAMES DEWHARK
		Z Z Z	9.46 9.66 MMR		186 MEST84 1 CORR77	UNITED STATES RETHERLANDS
						ISRAEL ISRAEL
		rag S			PAGESZ 1 CORR77 321 MEST84	
		er.	2000 2000 2000 2000		ZOETA HESTA	METHERLANDS UNITED STATES
		E E	. 75E-0		MEST8	UNITED STATES, NEW YORK

CASIB	SUBSTANCE		REPORTED	SCREENING	* REFERENCE	PLACE
		CGN	CENTRATIONS	! ! ! !	SAMP	
67563 METHANE, TRICHLORO-	TRICHLORO-	NO NO	-00	8.00E-01	1 SAUES1	GULF OF MEXICO
		E.	20		17 KAIS83	HEH YORK
		5X.	. 00th		SAUES 1 TRUSS	GULF OF MEXICO
		K I	.30E-0		KAISS	CAHADA, GNTARIO
		3	. 00E-0		PEAR	ENGLAID, THAMES RIVER
			. 10E-0		34 CHAR83)
		:5	. 80E-0			JUNEAU PRINCE
		¥	. 70E-0		1 VAIIR52	SOUTH AFRICA
		, i	. 40E-0		TRESSO PFIER	CHIMA
		3	. 90E-0		1 TRUS80	PETLIPPINES
		52	. 00E-0		ZOETSO 1 TRIICED	METHERLANDS FIG. ASD
		ž	-40E-0		TRUSSO	PERU
		Ę	0-1100 0-1100			THAILAND
		3	. 04E-0		PAGESZ	CZZIED SIATES
		E F	.10E-0		1 180580	AUSTRALIA
		ž	30E-6		T VANER	NETHERLARDS AGITH AFBICA
		3	. 40E-0		Y VAIIR82	SOUTH AFRICA
		Š				BRAZIL Sente
		Z.	. 00E-0		1 VAIIRBZ	SOUTH AFRICA
		Ž	10E-0		1 VANRAS	SOUTH AFRICA
		S	. 50E-0		1 VANERS	AUCUST AFKICA
		Z.	. 70E-0		18 CHAIRS	FRANCE
		; ; ;	. 20E-0		VANRE	STITH AFRICA
		<u> </u>	. 40E-0		VANRA	SOUTH AFRICA
		ž	296-0		CHARIS	子に入れてた。
		Z.	.70E-0		TRUSSO	VENEZUELA
		25.4	.//E-0			BRAZIL
		E	.70E-0		•	HELHEKLANDS URITED STATES - MEM YO
		ž ž	.59E-		DEFESA.	
		2	. 00E-0			UNITED STATES
			-40E-		0	UNITED STATES
		Z.C.			32 SCHR81	KEH YORK
		I.	. 49E-		aw	UNITED STATES
			. 30E-	•	100	CHITED STATES
			300.		6	REM YORK

Table 5. (Continued)

Table 5. (Continued)

CASID		SUBSTANCE	TYPE	REPORTED	CREE	* REFERENCE	PLACE
			SOS	ENTRATIONS	MG/L	SAFF	
88062 PHENOL,		2,4,6-TRICHLORG-	Y Y Y Y	0007	8.20E-61	MEGR79 HEGR79 HEGR19	NETHERLANDS HETHER ANDS HETHER ANDS HETHER ANDS
			n n n n	70E		HEGH79	NETHERLANDS NETHERLANDS NETHERLANDS HETHERLANDS
120832 PHENOL,	HENOL,	2,4-DICHLORG-	, K. K.	-65	5.80E-01	ИЕСН79 НЕСН79 НЕС179	NETHERLANDS NETHERLANDS NETHERLANDS
583788 PH	HENDT.	PHENOL, 2,5-DICHLORO-	, K		NO VALUE	MEG1179 MEG1179	netherlands Hetherlands
87650 PH	PHENGL,	2,6-DICHLORG-	555		2.94E+00	MEGH79 MEGH79 HFG=174	METHERLANDS HETHERLANDS HETHED AND
			.	e cost		HEGITO HEGITO HEGITO	NETHER ANDS NETHER LANDS NETHER LANDS
			ZZZ WW.			11EG779	HETHERLANDS NETHERLANDS NETHERLANDS
95578 PH 95772 PH	PHENOL:	2-CHLORG- 3,4-Dichlorg-	XX:		4.40E-01	MEG179 MEG179	NETHERLANDS NETHERLANDS NETHERLANDS
106489 PH	PHENOL,	4-CHLORG-	22.	-20E	2.61E-01	MEGH79 MEGH79	NETHERLAIDS NETHERLAIDS
108430 PH 78875 PR	PHENOL PROPANÉ,	3-CHLORO- 1,2-DICHLORG-	E E E		5.70E-01 8.60E-01	MEGM7 MEGH7 KAIS8	METHERLANDS METHERLANDS MEH YORK
				4.2.4. mmod		KAISS HESTS HESTS	ONTA
108883 TOLUENE	DL VENE			5	5.00E+00	56 HEST8 66 HEST8 1 SAUE7 SAUE7	TATES TATES HEXIC
			555	0.00		SAUES SAUES SAUES	TEXT TOUC
			555	24. 100. 100.		1 SAUE78 1 SAUE78 1 SAUE78	TEXT TO COL
			5555	20.4 DA71.		SAUEZ SAUES SAUEZ SAUEZ	GULF OF MEXICO GULF OF MEXICO GULF OF MEXICO GULF OF MEXICO
			e se se	2000 3000		1 TRUSBO 20 ZGETBO 1 SAUE78 1 TRUSBO	
			i 3				⋖

Table 5. (Continued)

CASID	CASID SUBSTANCE	TYPE	REPORTED	SCREENING	# REFERENCE	PLACE	
0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0		CONCENTRATIONS MG/L	HG/L	SAME		
		nnegerennie eere			232 282 282 292 1 28 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	PERU CHINE CHINE CHITED STATES UNITED STATES VEHEZUES NETHERLAND EGYPT PAZIL PHILIPPINES UNITED STATES UNITED STATES UNITED STATES UNITED STATES	
		35	3.00E-01		TAMBS:	HETHERLANDS	

Screening concentration is based on LD50 data. Note:

* FDW = finished drinking water. GW = ground water. SW = surface water. OW = ocean water.

Cyclopentadiene, hexachloro-

The only concentration value we were able to obtain was above the organoleptic threshold.

Ethene, tetrachloro-

Its occurrence is worldwide. The highest concentration was for a U.S. surface water. The other values were below the screening concentration.

Methane, trichloro-

The highest concentration values are from the U.S., but worldwide values are not significantly lower.

Phenol

All but one of the values listed in our basic source data, Table D-1, Appendix D, are from Norway. Organoleptic response would be zero at these concentrations; however, we did find higher values in the WaterDROP data base for waters in the U.S. The highest was 60 mg/L at the outlet of an industrial discharge. This value would definitely be above the taste/odor threshold. (WaterDROP data, as explained previously, were used for comparison.)

Phenois, chloro-(mono-, di-, trichloro-) Most of the values were from a study of the Rhine River in the Netherlands.

Phenol, pentachloro-

All surface-water monitoring data we surveyed were from the Netherlands or Japan. The largest concentration in WaterDROP is an order of magnitude higher than our survey data.

Propane, 1,2-dichloro-

All values are for tap waters in North America.

Toluene

It is found worldwide, but the problem concentration was for a landfill leachate in the Netherlands.

We can place most of the compounds with ratios less than 0.1 in Table 4 into two classes: phenols and their chlorinated homologs and chlorinated hydrocarbons. The compounds with ratios of 0.1 to 1.0 are also primarily chlorinated hydrocarbons and phenols. Tetrachloroethene, 1,2-dichloropropane, trichloroethene, dichloromethane, and dibromomethane are found in many of the same studies. Toluene (i.e., methylbenzene) and

ethylbenzene are found in the same studies. Likewise, all chlorinated phenols are found together. We would expect the total olfactory contributions of these combinations to be greater than the single contributions.

The compounds with the greatest worldwide distributions (see Table 5) are ethylbenzene, tetrachloroethene, trichloromethane, and toluene. Of these four substances, the one with the highest screening ratio was trichloromethane, with a ratio of 4.3. It was followed by ethylbenzene (3.0), toluene (2.5), and tetrachloroethene (1.25). None of the measured concentrations is greatly above the detection thresholds, and therefore these substances may not actually cause adverse organoleptic responses. Our data on the chlorinated phenols were derived from a single study of the Rhine River; however, phenolic compounds can be expected in other similar rivers receiving industrial discharges (we will discuss the phenolic compounds again in a later section).

An important source of uncertainty in the screening for potential organoleptic effects has been the lack of adequate psychometric data on the kinds of responses expected at the contaminant concentrations presented. If the threshold concentrations we have used for screening are far below levels that would actually cause a soldier to refuse water, then none of these substances truly represents a problem. To rectify this situation, we recommend that taste panels comprised of soldiers be used to quantify the behavioral responses to varying concentrations of the organic solutes noted above.

PREDICTED ORGANIC-SOLUTE CONCENTRATION MAXIMA IN THE SURFACE WATERS OF REPRESENTATIVE FOREIGN COUNTRIES

We recognize the fact that measurement bias may play a role in the screening comparisons we performed using observed-maximum concentrations. Therefore, to compensate for such possible bias, we augmented our initial screening process with one that uses predicted concentrations of the following organic substances:

- Acenaphthene,
- Acrolein,
- Acrylonitrile,
- · Chloromethane.
- Cyanide.

- 2,4-Dinitrophenol,
- Oil and grease,
- Pentachlorophenol,
- Phenol, and
- Phenol (4AAP)a.

These are major contaminants that are likely to be present in the waste waters from 13 major industrial sources in foreign countries. The industrial sources include: (1) aluminum forming, (2) coal mining, (3) copper forming, (4) foundries, (5) iron and steel mills, (6) leather tanning, (7) metal finishing, (8) nonferrous metal production, (9) ore mining, (10) organic chemical/plastic production, (11) petroleum refining, (12) pulp and paper production, and (13) textile manufacturing.

To predict the maximum concentrations in surface water of the organic substances listed above, we used a model that combines data for (1) the annual release of each of the organic chemicals into wastestreams by the U.S. industries that are in the previously mentioned 13 industrial categories, with data for (2) the annual production of the manufactured/mined material from each of the 13 industrial categories in five representative foreign countries and in the U.S., and (3) country-specific hydrologic data. Comparison between these predicted maximum concentrations in surface water and the corresponding estimates of threshold concentrations for toxic effects (based on 70-kg military personnel consuming water at a rate up to 15 L/d) or organoleptic responses will augment the screening for the set of organic compounds that could pose a human health risk to military personnel consuming field water in foreign countries.

Our approach for estimating the concentration of an arbitrary chemical in the surface water of a given country is based on the influence diagram shown in Fig. 4. This figure is based in part on the concepts described by Mackay. For our purposes, for example, the concentration of an organic compound in a river, lake, or estuary is related to (1) the amount of the compound in industrial effluents (both direct and indirect), (2) the advection (e.g., horizontal transport) properties for receiving waters, and (3) the attenuation of the organic compound's concentration in water by means of biogeochemical processes. The contribution of industrial effluents to organic-chemical pollution of surface waters is estimated by considering (1) the types of industries that release the compound into wastestreams, (2) the most prominent organic chemicals released into the wastestreams (per unit of production) by each industry, and (3) the level of output of

^aThe designation 4AAP refers to the method of detection, which involves the reaction of phenolic compounds with 4-aminoantipyrine (i.e., 4AAP).

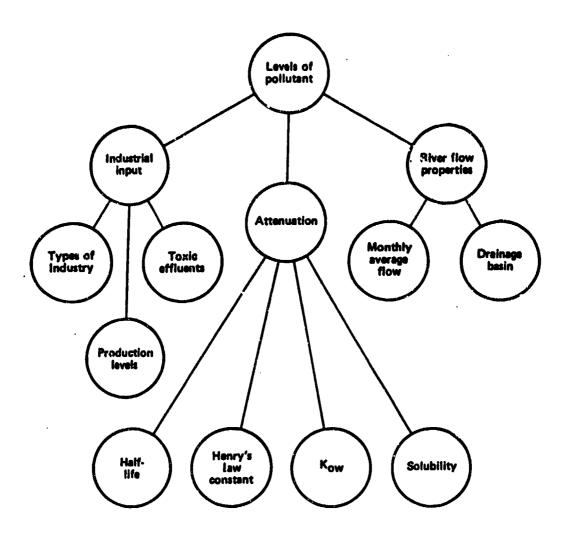


Figure 4. Influence diagram for processes that affect the concentration of organic solutes in surface waters.

manufactured/mined material each industry produces. The volume of surface water available for diluting the released organic chemicals is determined using river-flow properties (e.g., average-flow volume monthly). The physicochemical properties of the organic compounds released provide information on the potential attenuation of concentration by chemical decay (biodegradation, photolysis, hydrolysis, oxidation, and volatilization), sorption by sediments, and uptake by biomass. However, in order to simplify our calculation and provide a conservative upper bound for our concentration estimate, attenuation as a result of these chemical processes was excluded from our model. Dilution by advection within the water system, however, was included in the model. Figure 5 provides an illustration of a national water system as it is viewed for the purposes of this model.

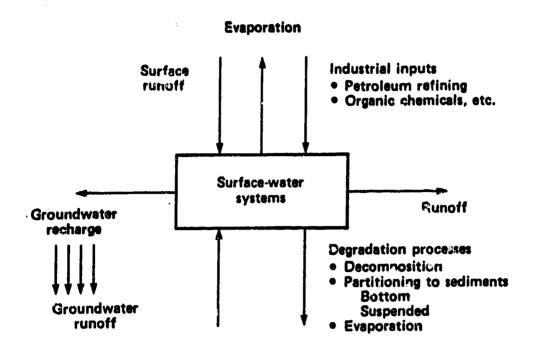


Figure 5. A schematic view of a national water system for use in screening organic solutes likely to be at potentially toxic levels.

The screening methodology was applied to each of the five representative countries according to the following mathematical expression:

$$C_{i,m} = \frac{1}{R_m} \sum_{j=1}^m F_{i,j} \cdot P_{j,m},$$

where

- c_{i,m} the concentration of chemical compound, i (g/L), in the surface waters of country, m;
- R_m = the discharge of the major rivers (L/y) in country, m, that are likely to be receiving industrial wastestreams;
- F_{i,j} = the amount of chemical, i (g/y), contained in the raw-effluent wastestream generated annually by industry, j, in the U.S. (The hoice of raw-effluent wastestream instead of treated-effluent wastestream tends to provide added conservatism to this figure; this assumption also is realistic for most developing countries); and

P_{j,m} the ratio between the annual manufacturing output produced nationally for industry, j, in country, m, and the same production-output value for industry, j, in the U.S. We define this ratio as the production-output factor (dimensionless) for industry, j, in country, m.

Statistics compiled by the United Nations 14 concerning the annual manufacturing output produced nationally in a particular country by a specific industrial category were used to derive the production-output factor $(P_{j,m})$. Data on the amounts of the organic chemicals of interest contained in the raw-effluent wastestreams generated annually by an industry category in the U.S. (i.e., $F_{i,j}$) were obtained from a U.S. EPA-sponsored study concerning the discharges of 86 organic contaminants released from 24 industries in the U.S. 15 The surface-water volume available in each country (R_m) for dilution of the organic chemicals contained in the raw-effluent wastestreams generated annually by the combination of all 13 industrial categories was obtained from a reference work edited by van der Leeden 16 ; these figures correspond to the volumes of major surface-water systems in each country that are likely to receive industrial wastestreams, and therefore they are not necessarily equal to the total surface runoff in each country.

Once the estimated concentration of a specific organic chemical is calculated for a particular representative country $(C_{i,m})$, based on the industrial mix and the receiving-water flow, the chemicals are ranked for each country, using the screening-criteria formula:

$$S_{i,m} = \frac{C_{i,m}}{T_i} ,$$

where

Si _ = screening-criteria ratio (dimensionless) for chemical, i, in country, m,

C_{i,m} - concentration of chemical compound, i (mg/L), in the surface waters of country, m, and

an estimate of the threshold concentration (mg/L) in water of chemical, i, for toxicity or organoleptic effects, based on 70-kg military personnel consuming water at a consumption rate of up to 15 L/d.

Table 6. Production-output factors^a for estimating source terms from the 13 industrial/mining categories that we consider to be major sources of organic-chemical pollution in foreign countries.

			Country		
Industry	Egypt	Iran	Israel	Thailand	Turkey
		Pro	duction fact	ors	
Aluminum forming	2.00E-03	1.60E-03	0	0	8.30E-03
Coal mining	0	1.60E-03	0	9.80E-04	7.50E-03
Copper forming	· 0	0	0	0	1.10E-02
Foundries	7.20E-03	8.13E-05	8.10E-04	2.60E-03	1.50E-02
Iron and steel	7.50E-03	0	9.90E-04	3.20E-03	1.70E-02
Leather tanning	0	4.50E-02	0	0	1.10E-01
Metal finishing	1.80E-02	3.80E-02	3.90E-03	2.80E-03	2.60E-02
Nonferrous metals	0	0	0	0	4.30E-03
Ore mining	1.40E-02	7.20E-03	1.60E-05	1.30E-03	4.50E-02
Organics/plastics	4.90E-04	4.70E-05	1.30E-03	3.90E-06	5.80E-03
Petroleum refining	1.80E-02	3.40E-02	1.40E-02	1.10E-02	2.00E-02
Pulp and paper	1.20E-03	7.80E-04	1.10E-03	4.60E-03	6.80E-03
Textiles	0	2.60E-02	0	1.50E-01	4.00E-02

a Production-output factors are calculated by dividing the annual manufacturing or mining output from an industrial/mining category for a foreign country by the same value for the equivalent industrial/mining category in the U.S. The amount of organic-chemical pollutant contained in the raw-effluent wastestreams for all 13 industrial/mining categories in a foreign country is then the sum of the product of the production-output factor and the amount of organic-chemical pollutant in the raw-effluent wastestreams generated by each industrial/mining category in the U.S.

Application of the Model to Five Representative Countries

The model described above was applied to the following countries: Egypt, Iran, Israel, Thailand, and Turkey. These countries were selected because they possess the wide range of industrial and climatic conditions that are most likely to be encountered in foreign countries. Table 6 lists the production-output factors for the thirteen industrial categories that we used to scale the level of organic-chemical pollutant contained in the raw-effluent wastestreams generated by these industries in the five foreign countries just mentioned. Table 7 lists the amounts of the pollutants contained in the raw-effluent wastestreams generated annually in the U.S. by each of the 13 industrial categories. Table 8 presents the average and low-dilution volumes provided annually by major rivers

Table 7. U.S. production of organic by-products in g/y for each industrial category.

Industry	Acenaph- thene	Acrolein	Acrylo- nitrile	Chloro- methane Cyanide	Cyanide	2,4- Dinitro- phenol	Oil and grease	Penta- chloro- phenol	Phenol	Phenoi (4AAP)a
Aluminum forming					1.16E+06		8.58E+08		8.65E+05	
Coal mining					2.66E+06					8.87E+07
Copper forming					2.80E+06		4.59E+11			2.13E+08
Foundries	1.43E+07					1.10E+06	3.13E+10	1.14E+07	2.55E+09	3.44E+09
Iron and steel			5.46E+07				2.93E+11	6.00E+06	1.80E+10	1.39E+10
Leather tanning	7.43E+05				2.39E+06		1.98E+10	6.01E+06	1.60E+08	
Metal finishing					1.43E+10		1.42E+11			
Nonferrous metals	2.25E+07				1.21E+08		1.17E+09		2.51E+08	2.52E+08
Ore mining					6.67E+07					5.66E+06
Organics/plastics	5.24E+09	1.04E+10	6.89E+09	1.87E+08	1.87E+08 1.79E+09	5.45E+07		7.85E+06	1.77E+10	
Petroleum refining	8.14E+07	3.02E+05			8.41E+07	4.80E+08	6.49E+10	9.48E+05	5.90E+08	9.70E+09
Pulp and paper								4.87E+07	2.12E+08	
Textiles			1.85E+07		1.37E+07		3.73E+10	1.23E+07	2.01E+07	

^a The designation 4AAP refers to the method of detection, which involves the reaction of phenolic compounds with 4-aminoantipyrine (i.e., 4AAP).

Table 8. Annual average and low dilution volumes provided by major rivers in each of the candidate countries.

	Dilu	tion volume ^a (L/y)
Country	Low	Average
Egypt	2.21E+13	8.20E+13
Iran	3.15E+11	9.46E+12
Israel	9.10E+10	1.96E+12
Thailand	5.20E+13	2.90E+14
Turkey	1.40E+13	4.00E+13

^a The low dilution volume was calculated by multiplying the lowest monthly flow by 12.

Table 9. Annual mass discharges of the major organic contaminants released into the raw-effluent wastestreams of all 13 industrial/mining categories in each of five representative countries.

		····	Country		
Contaminant	Egypt	Iran	Israel	Thailand	Turkey
		Mass	lischarged, k	g/y	
Acenaphthene	4.13E+06	3.05E+06	7.96E+06	9.53E+05	3.24E+07
Acrolein	5.08E+06	4.97E+05	1.35E+07	4.37E+04	6.01E+07
Acrylonitrile	3.79E+06	8.06E+05	9.02E+06	2.98E+06	4.16E+07
Chloromethane	9.14E+04	8.77E+03	2.43E+05	7.28E+02	1.08E+06
Cyanide	2.61E+08	5.47E+08	5.92E+07	4.31E+07	3.88E+08
2,4-Dinitrophenol	8.32E+06	1.57E+07	6.52E+06	5.07E+06	9.54E+06
Oil and grease	6.15E+09	9.48E+09	1.78E+09	7.73E+09	1.92E+10
Pentachlorophenol	2.07E+05	6.68E+05	9.22E+04	2.17E+06	1.83E+06
Phenol	1.72E+08	2.90E+07	5.13E+07	7.46E+07	4.79E+08
Phenol (4AAP) ^a	3.04E+08	3.30E+08	1.52E+08	1.60E+08	4.86E+08

a Phenol (4AAP) includes soluble phenolic compounds.

in each of the five representative countries. Table 9 contains the estimated total amount of each of the organic-chemical pollutants generated in each country from the combination of all 13 industrial categories. An estimate of the maximum and mean concentrations for each of the pollutants of interest in the surface waters of each of the five representative countries are presented in Table 10. Finally, Table 11 contains the values representing toxicity or organoleptic threshold concentrations (T_i) and screening-criteria ratios $(S_{i,m})$ for each of the organic chemicals of interest.

According to the screening-criteria ratios presented in Table 11, dissolved oil and grease is of potential concern in virtually all foreign countries. The screening-criteria value for oil and grease was computed using the Soviet taste/odor standard of 0.1 mg/L for crude oil (high sulfur) and kerosene reported by Bedding et al. ¹⁸ The importance of oil and grease is not surprising because oil and grease are major components of many industrial effluents prior to wastewater treatment. We were able to obtain some water-quality data that corroborate this prediction from our screening. For example, Moursy measured concentrations of oil and grease in the Nile River that were typically between 1 and 30 mg/L. The highest of these concentrations were recorded near the major industrialized areas along the river. In another study Cordero et al. ²⁰ sampled water from the Bermudez River in Costa Rica at three separate locations and measured oil and grease levels ranging from approximately 30 to 50 mg/L. This river also received effluents from various industrial sources.

In addition to oil and grease posing potential problems for military personnel from an organoleptic standpoint, these hydrocarbons can interfere with the operation of the Reverse Osmosis Water Purification Unit (ROWPU). Specifically, oil and grease can create a film on the reverse osmosis membrane and thus decrease its ability to desalinate water.

The other organic chemicals of potentially greatest concern to U.S. military personnel in foreign countries are the phenolic compounds detected by their reaction with 4-aminoantipyrine (i.e., 4AAP). The screening-criteria values are all greater than one for this class of organic compounds for each of the five representative countries. These phenolic compounds are not a problem in drinking water from a toxicological standpoint because the WHO limit for 2,4,6-trichlorophenol is an organoleptic limit.

Generally, Israel is the country predicted to have the highest concentrations in their surface waters (see Table 10) of the organic chemicals of interest. However, the predicted concentrations that appear in Table 10 are very conservative because we have assumed that there is no pretreatment (e.g., chlorination) of raw-effluent wastestreams, and we have used the low-dilution volumes for receiving waters.

Table 10. The estimated maximum^a and mean concentrations of each of the major organic chemicals in the surface waters of each of five representative countries.

					Country	ý				
	<u>a</u>	Egypt	Iran	_	Israel	-	Theiland	9	Turkey	4
	Max.	Mean	Мах.	Mean	Max.	Mean	Max.	Mean	Max.	Mean
Contaminant					Concentra	Concentration (mg/L)				
Acenaphthene	1.87E-04	1.87E-04 5.04E-05	9.67E-03	3.22E-04	8.76E-02	4.06E-03		1.83E-06 3.29E-08 2.31E-03	2.31E-03	R. 10E-04
Acrolein	2.30E-04	2.30E-04 6.19E-05	1.58E-03	5.25E-05	1.48E-01	6.87E-03	8.40E-07	1.51E-07	4.29F_03	1.502.03
Acrylonitrile	1.71E-04	1.71E-04 4.62E-05 2.E	2.55E-03	8.52E-05	9.93E-02	4.60E-03	5.73E-05	1.03E-06	2.97F-03	1.04P_03
Chloromethane	4.14E-06	4.14E-06 1.11E-06	2.78E-05	9.27E-07	2.67E-03	1.24E-04	1.40E-08	2.51E_09	7.738-06	2 700 cis
Cyanide	1.18E-02	3.18E-03	1.73E+00	5.78E-02	6.52E-01	3.02E-02	8.29E-04	1.49E-04	2.77E-02	9.706.03
2,4-Dinitrophenol	3.77E-04	1.01E-04	4.96E-02	1.66E-03	7.18E-02	3.33E-03	9.75E-06	1.75E-06	6.82E-04	2.398-04
Oil and grease	2.78E-01	7.50E-02	3.00E+01	1.00E+00	1.96E+01	9.07E-01	1.49E-01	2.66E-02	1.37E+00	4.79E-01
Pentachloropherol	9.35E-06	2.52E-06	2.12E-03	7.06E-05	1.02E-03	4.71E-05	4.17E-05	7.47E-06	1.31E-04	4.58E-05
Phenol	7.81E-03	2.10E-03	9.18E-02	3.06E-03	5.65E-01	2.62E-02	1.43E-03	2.57E-04	3.42E-02	1.20E-02
Phenol (4AAP)	1.37E-02	3.70E-03	1.05E+00	3.49E-02	1.68E+00	7.77E-02	3.08E-03	5.52E-04	3.47E-02	1.21E-02

^a The maximum concentration is calculated from a dilution volume by multiplying the lowest monthly flow volume by 12. The mean is calculated from annual average flows.

b Phenol (4AAP) includes soluble phenolic compounds.

Table 11. Toxicity- or organoleptic-threshold concentrations in water (T_i) , based on 70-kg military personnel consuming up to 15 L/d of drinking water) and screening-criteria ratios $(S_{i,m})$, calculated by dividing the estimated maximum concentration in the water of a particular country, $C_{i,m}$, by T_i) for each of the major organic chemicals (i) in each of five representative countries (m).

	Screening			Country		
	concentration	n Egypt	Iran	Israel	Thailan	
Contaminant	(mg/L)			Concen	tration ratio) 8
Acenaphthene	2.00E-02 ^a	9.35E-03	4.83E-01	4.38E+00	9.17E-04	1.16E-01
Acrolein	3.20E-01 ^a	7.18E-04	4.92E-03	4.63E-01	2.63E-06	1.34E-02
Acrylonitrile	2.00E+00 ^a	8.57E-05	1.28E-03	4.96E-02	2.87E-05	1.49E-03
Chloromethane	2.00E-03 ^a	2.07E-03	1.39E-02	1.34E+00	7.00E-06	3.86E-02
Cyanide	2.00E+00 ^b	5.90E-03	8.65E-01	3.26E-01	4.14E-04	1.38E-02
2,4-Dinitrophenol	3.00E-02 ^a	1.26E-02	1.65E+00	2.39E+00	3.25E-03	2.27E-02
Oil and grease	1.00E-01 ^C	2.78E+00	3.00E+02	1.96E+02	1.49E+00	1.37E+01
Pentachlorophenol	5.00E-03 ^d	1.87E-03	4.24E-01	2.04E-01	8.34E-03	2.62E-02
Phenol	3.50E+00 ^a	2.23E-03	3.62E-02	1.61E-01	4.10E-04	9.77E-03
Phenol (4AAP) ^e	1.00E-04 ^f	1.38E+02	1.04E+04	1.68E+04	3.08E+01	3.47E+02

a Lowest health standard or guideline (see Appendix A).

BIOCHEMICAL COMPOUNDS RELEASED BY ALGAE AND ASSOCIATED MICROORGANISMS

The ubiquitous aquatic microorganisms, cyanobacteria (blue-green algae) and actinomycetes (Gram-positive filamenteus bacteria that grow in close association with cyanobacteria), have been identified as the source of taste- and odor-producing biochemical compounds in surface waters, particularly drinking-water reservoirs (Krasner et al., 1981). Furthermore, there is even circumstantial evidence suggesting that a causal relationship exists between otherwise unexplainable outbreaks of adverse health effects in human populations and the presence of toxic biochemicals released by

b Interim value recommended by Scofield et al. for field-water supplies. 17

^C Taste/odor standards of USSR for oil and kerosene. ¹⁸

d Value calculated from oral LD50 (see Table 2).

e Phenol (4AAP) includes soluble phenolic compounds.

f WHO limit for 2,4,6-trichlorophenol (see Appendix A).

cyanobacteria into public drinking-water supplies. 22-27 Therefore, in addition to the organic chemicals previously discussed, we also assessed the significance of these natural organic compounds from the perspective of military field-water quality.

Our assessment revealed that there are two taste- and odor-causing metabolites of cyanobacteria and actinomycetes--geosmin and 2-methylisoborneol (MIB)--that can be of military concern in field water, especially when algal blooms are present. There are two reasons that justify this conclusion. First, these metabolites can occur in field water at concentrations above organoleptic thresholds. For example, according to a survey cited by Zoeteman (1981)²⁸ comparing the maximum concentration measured in drinking water of taste-impairing substances with their odor-threshold concentration (defined as the concentration at which 50% of the subjects can detect the odor of the substance), the ratios of maximum concentration to odor-threshold concentration were among the greatest for geosmin and MIB (i.e., 1.5 = 0.03 µg/L of water/0.02 µg/L of water). Second, these biologically released substances resist oxidation and therefore are difficult to remove by standard water-treatment processes such as chlorination (Krasner et al., 1981).²¹ Consequently, military personnel exposed to field water containing maximum concentrations of geosmin and MIB may refuse to consume such water based on objectionable organoletpic properties (i.e., taste and odor) and therefore can become susceptible to the performance-degrading health effects associated with dehydration. Moreover, chlorination, which is the water-treatment technique most commonly available in the field, normally is ineffective in destroying these compounds.

Another group of biochemicals that may be of military concern are the alkaloid, lipopolysaccharide, and polypeptide metabolites that are released by cyanobacteria. These natural organic substances have been associated with the death of animals, especially cattle, and are implicated as being toxic to man (Carmichael, 1981). Consequently, the potential for performance-degrading effects from human exposure to these substances should be evaluated further.

SUMMARY AND RECOMMENDATIONS

Our screening effort has addressed substances that could potentially cause toxic or organoleptic effects in troops that drink field water. To minimize the omission of substances that might adversely affect troops, we made conservative assumptions for screening for those organic chemicals that could be present in field water at concentrations that could produce organoleptic effects or toxicity in military personnel forced to consume up to 15 L/d of field water. Moreover, we assumed that no treatment

would occur prior to water consumption. From our comparison between reported maximum-observed concentrations and threshold concentrations for toxicity (based on a 15-L/d consumption rate for a 70-kg individual), we conclude that there is a very low probability that organic solutes in field water will cause direct, debilitating effects in troops. However, this assumes that troops follow existing doctrine regarding the placement of water-supply points. Siting a water point directly below a sewage outfall on a stream or river, for example, greatly enhances the risk of health effects. Likewise, the use of a well that is in the immediate vicinity of actual or possible surface industrial contamination poses an increased health risk. Avoiding these obvious situations is a key precaution in the management of health risks from all contaminants of field water.

We have identified several compounds that have the potential for causing organoleptic effects (e.g., objectionable taste or odor). Among the most important compounds in this group are trichloromethane, ethylbenzene, toluene, and tetrachloroethene because they have the greatest worldwide distribution. In addition, chlorinated phenols, as well as oil and grease, could impair the potability of field water. Our screening for organoleptic effects, however, utilized concentrations for taste—and odor-detection thresholds instead of concentrations related to a behavioral response, such as refusal to drink poor-tasting water. We therefore recommend that taste panels comprised of groups of soldiers be used to quantify the relationship between the concentrations of these substances, an organoleptic property, and various behavioral responses (see, for example, Daniels and Layton, 1988³⁰). This research could provide a data base that would support a more definitive analysis for determining the organic chemicals likely to impair the potability of field water. An important issue that has emerged is the effect that oil and grease could have on the operation of a ROWPU. Studies are needed to determine the concentrations of oil and grease that could impair the efficiency of reverse osmosis membranes.

Finally, we indicate that compounds released into water by the aquatic microorganisms, cyanobacteria and actinomycetes, can also be of particular military concern. These substances fall into two categories: (1) those that impair the taste and odor of drinking water and (2) those that could produce toxic health effects following ingestion or nonconsumptive exposure. Geosmin and MIB fall into the first category and alkaloid, lipopolysaccharide, and polypeptide toxins belong to the second one. Because of the potential for algal blooms in surface waters, we recommend that available data on the organoleptic and toxic properties of these metabolites be evaluated carefully so that consideration can be given to developing criteria and recommendations for their standards in field water.

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APPENDIX A

HEALTH STANDARDS AND GUIDELINES FOR ORGANIC WATER CONTAMINANTS

As a starting point for our screening analyses of organic contaminants that could pose health problems, we compiled relevant health standards and guidelines developed by domestic as well as foreign organizations. Table A-1 summarizes the results of our regulatory review. Included in the table are short-term (approximately 10 d or less) and long-term (chronic exposure) standards. For example, a 7-d standard is expressed as 7d, followed by a slash and the standard in mg/L (e.g., 7d/0.03). We have also listed substances that have been identified as hazardous by different organizations. For a number of substances we calculated an equivalent drinking-water concentration from Acceptable Daily Intakes (ADIs).

Table A-1. Regulatory status of organic compounds (excluding pesticides).

CASIO	Substance	Formula	2 2	EFA advis*	MAS SMAT 1*	HAS 341	std std	Soviet	tos tos	Other
							L/8			
83329 208968 141786	acemaphtheme acemaphthyleme acetic acid, ethyl ester	C12410 C1248 C44802	××			A 1433			20.0	
7320 108054	viet.	C3M602 C4M602						0.1		
107028 19961	acrolein acrylanide	C3#46	×						0.32	
79167 161791 1693111	acrylic acid acrylemitrile adipentirile	C3#402 C3#38 C6#882	×	•						78
9962		CEMECTA CEMENTOS						0.7		2
2 1 2 2 2 2 2 3 2 3 3 3 3 3 3 3 3 3 3 3	aniline, e-nitre- aciline, p-chlore- aniline, p-nitre- anisole	CENEM202 CENECTH CENEM202 C7MB0						0.5		222
11432 100907 100414	benzene benzene, chloro- benzene, ethyl-	25.00 C. C. C	×××	104/0.23 74/12.6	74/12.6	•	0.00	6 	0.62	
118741 90953 577599 600935		C6C16 C6MSND2 C12M18 C6MC15	. * *			0.035	0.00001	6.05	. 6. 6. 59. 6	5
25.0000 25.50 25.900 25.900 25.173		C6H4C1W02 C6H4C1Z C6H4C1Z C6H2C14 C6H2C14	* *				0.0003	0.05	-0.03 -0.03	
\$ 5 5 5 5 5 5 5 5 5 5 5 5 5 5	perzee, 1,4-dichlorg- berzielne, 2,3'-dichlorg- benze(a)anthracese	C6MC12 C12M12M2 C12M10C12M2 C16M12	* * * *			6.43	0.00		6. 14	
2020 2020 201242 20168	ben 20[a]pyrene ben 20[b]fluoranthe.e ben 20[ghl]perylene ben 20[h]fluoranthene	C228112 C228112 C228112 C208112	***	104/0.025			0.00001			
87678 87696 87696 87696	Dipmenyl butachlor butadlene, bezachlora- buten-2-ode	C12N10 C17N26C1N02 C4C16 C4N80	×	14/7.5		2.1 0.35			. .	2 Z Z

Table A-i. (Continued).

CASIO	Sebstance	Formula	2 2	EPA advis*	MAS saeri*	MAS 1 ped	975	Soviet std	\$ 15 S	Other
							// ba			
218019	-	C10M12 C7K7C10	ны							
~ 4	cresol-a, 6-chlora-	C7W7C10								
1321104	cresol, chlere-	C7W7C10							₩ 6 	
534521	cresel, 4,6-diattre-	C786M205	×						0.0134	
2	crotesitrile	CANSA						0.7)))	
1002	Cyc) obexane							S - C		
37	cyclohexane, nitre-	CEMSM02								
	4	C6#120						6.3		
1.783	cyclobexene extent							 6		
17474	cyclepentatione, hexacklare-	55MC36	×					9.00	0.0	
12/24	di-e-propyinitrosaniae	C6814820	×							
53763	dibenzo[a,b]anthracene	C22H14	×							Ž
19811	dibutyl tin dilauente	C32H64045m	•					0.1		
X :	dichierebydrie	C3#6C12						S. 0		
F 24	dictayl effer malegic dictayl tin dicarrists	CENTE			•			 		
109097	diethylamine	C4k1)#								
55165	diethylamine, M-nitroso-	C4H10M20						2	0.000000	נאנ
	disapravlasia									
2	dimethyl phonyl carbinel	C9#120						. 6 . 6		
4275	d Verethy lamine	C2171	:					9.1	,	
25136554	dioxane, disethyl-	202 LESS	×						0.0000014	4 car
1746016	diexia	C12N4C1402	×							
	dipterylatives. n-nitroso- dipterylativesamine		×						976	5
75003		CZHSCI	×				0.003			
75343	ethane, hexachloro- ethane, 1,1-4tchloro-	C2C16	×							5
11556		C2NDC13	4 ×	6/1.0	14/0.49				3.4	
2000K	defende 1 1 3 design	20000	;		10.6/201					
335	ethane, 1,1,2,2-tetrachlere-	C2K2C 14	× ×							25
107062	1.2-dichloro-	CZHAC12	: ×				6.9			2 2
1000	othere, chiero- (viny) chlori	4e)C2H3C3	×				•			}
	-	5	×	104/0.17	104/0.175 34/0.024					5
				c/0.05						

Table A-1. (Continued).

79016 etheme, tricilator—	C.SIO	Sebstance	Formula	2 2	EPA advis*	MAS saerl*	35	120 std	Saviet std	3.5	S Per
								2			
### ### ### #### #####################	79016	etheme, trichloro-	CZMC13	×	16/2.	14/0.105		9.8			3
CHAPTON 1.1-dichlora-(trans) CRUZI2 X (20.07)	75354	ethene, 1,1-dichlora-	CZNZC12	×	14/1.0			0.000	-		3
### ### ### ##########################	540590	etheme, 1,1-dichiere- (trans)	CZNZC12	×	14/2.7						•
Chief Chie	542881	ether, bis(chloramethyl) ether, bis(2-chloroethyl)		××	194/0.27						•
Chicken Chic	10 86 0 101553	ether, bis(2-chlorolsopropyl) ether, 4-bromophenyl phenyl		××						0.0347	3 3
Fluoranthese C16410 x C45.	7005723 107211	ether, 4-chlorophenyl phenyl- ethylene glycel	C12H12C10 C2M602	×		14/19.					
furth formsleebyde (HZO 14/0.03 14/0.03 0.5 14/0.1	206440	fluorantheme fluorame	016410	× 1		c %				8.	
Furth Furt	50000 8247961	forms idebyde fuel oil #2	CN20	×	14/0.03				9.5		
Peptan=2-0	10000	furan	C4#40		104/0.1				•		
Nexachloro cyclobaxano, delta- Cidici	5235 1178 128	heptan-2-ol heptyl alcohol	C7#150						0.00		
Market 1970cyc 1984-16 x	319068	_	91.39893	×					0.003		5
December	319657		C6H6C16 C6H5C16	* *							3 3
Decided control of the control of	1000		C1386C1602	:			0.035				
hexan-2-one C7H140 hexante hexante C2H140 14/13. 5.0 5.0 5.0 6.01	100970	nexametry (ened taning) becamethy lenetetraning	CENTENZ CENTZNA					•	0.0		
hexanate	519786	bexan-2-one	C7H140						e.		2
hydrazine	3536 /58 110543	bexand bexand	C10H6C1603		14/13				9.0		Ì
Nydrazine, phenyl-	302012		H4H2	•	<u>:</u>				9		
Indemo[1,2,3-cd]pyrene	122667		CONGRE	:					0.0		
1socrotomitrile	193395	indeno[1,2,3-cd]pyrene	C22M12	< ×							
150ptorone	35	1socrotonitrile	CANSH						0.1		
kerosene	75310	1Soptorose 1Soprosylasies	C98740	×						5.5	
	143500	Kepone	C10C1100				S		2.0		
	90230	kerosene	C10-C16 HC		16/6.1		:				

Table A-1. (Continued).

CASIB	Substance	Form)	X &	estape advise	TAS SABT 1ª	SAT P	S 25	Soriet	Ş	Other
			1				l/6m			
75865 107277 62741 80626	lactuaitrile, 2-methyl- mercury, chloroethyl- mercury, diethyl- methacrylic acid, methyl ester methane, bis(2-thloroethoxy)-	C4H7ND C2H5C1Ng C4H10Ng C5HB02 C5H10C1202	×			3.5		0.00 0.000 0.000 0.000		·
76297 75274	bromo- bromodichlore-	CH36r CH8rc12							0.092	3
15092	methane, dichlora- methane, dichlora-	CHECIS		14/13	14/18. 14/35. 104/1.3	74/5.				555
\$6238	methame, dickloredifluore- methame, tetrachlore-	CC12F2 CC14	××	14/13.	14/350. 74/5.6 14/14.		6.003	6.9	ฑ่	5 5
75252 67663 75694 74895 90153 135193 91203	methame, tribromo- methame, trichlorofluoro- methame, trichlorofluoro- methylamine methylol m-acrylamide maphth-1-ol maphth-2-ol	CHSH3 CC13F CC13F CH5H CH5H C10HB0 C10HB0 C10HB0	***	14/22 14/22 14/80 74/8.	14/22. 14/80. 74/8.		.	- 0 0 0 - 0 0 - 0 - 0 0 - 0	3.2	55
1335071 22234131 1321648 42 1321659 91587 117006 924613 143008 165113	naphthalene, hexachloro- naphthalene, ectachloro- naphthalene, pentachloro- naphthalene, tetrachloro- naphthalene, 2-chloro- naphthalene, 2-chloro- naphthalene, 2-methyl- naphtho-1,4-quinene, 2,3-dichlo altresodi-e-butyliamine nomyl alcohol p-doxane			14/5.7	,			6.25 10.0	0.0000 0.0000 0.0015 0.0015 0.0000 0.0000 0.0000 0.0000	ä
106503 11097691 11109282 111111165 53969219 1287236	p-phenylenediamine pcb (4raclor 1254) pcb (araclor 1221) pcb (araclor 1221) pcb (araclor 1242) pcb (araclor 1248) pcb (araclor 1248)	C644842 C1245C15 C1249C1 C1249C13 C1244C14 C1244C16	XXXXXX							

Table A-1. (Continued).

CASIO	Substance	Formula	¥ &	EPA advis*	RAS Saar 14	3.5	Mic	Soviet std	to K	ge
							Vie 1			
136254	pentanate	(1)								
9 0105 9	phenanthrene	C14K10	×					5.5		
256001	Frenci	05450	: 34							
	phenol.	CEMSC10	ı						3.5	
19966	٠.	CENSN03						3 3	0.00G	
	_	CENSHOS	×							
	•	CENC 150	×					79.A		
95567 10030		CEMSC10	×							
80755	phenol, Cametry !-	C7860							6.669	-
•			×					90.0		
26305									0.0000	
70837	2.4-dichlors	CEMECIZO							٦:	
105679	~	CB#100	< >			•			p. 0003	
21262		CENANZOS	< ×						6.9	
45868		C6M3C130	:					9.63 6.63	0.0	
Ĭ	~	C6H3C130	×						9.0 0.0	Ž
2		C6H4C120							0.1	2
`	phone: C.0-dichioro-	C6#4C120								
106489	Dienol. 4-riblora-	COM4C120								
117617	Bithelate, bis (2-atteches)								. G	
19958	buty beary		× '		**	21.			15.	
84742	v		× ,						}	
2000	•	C2484204	. *						ž	
2956	_	C1241404	٠,							
13113 1886	porbalate, dimethyl	C10H1004	*						350.	
		CGM7H						\$6.0	313.	
1336363	Polychlerinated binhenyle							}	0.01	
		CICMICIE	×	16/0.125	1470.35				•	
25267156	polychloropinene	Closincia		104/0.025	17 /9.08					
2	Propose, chlorotrifiuoro-	CONTROL						6.5		
	propose, 1-chloro-2,3-epoxy-	COMPCIO			14/0.04			.		
21887	Propens. 1.2-41chlore.	Carre	1		74/0.53					
\$42756		COMCIO	ĸ						0.5	
									0.067	

Table A-1. (Continued).

CASIO	Substance	Formela	3 2	EPA advis*	MAS snari*	ENS ENS	Std	Seviet std	ings tox	other
							L/Ba			1 .
129000 110961 17 120003 121624 16009	pyrene pyridine, 2,5-dimethyl- pyrocatechol rdx (hexogam er cyclomite) resercinel	C16410 C5454 C7454 C64602 C3464602 C3464602 C54602	×		14/2.2			 	,	
39 643181 597648 100625 100625	sodium adipate stammane, dibutyldichlore- stammane, tetraethyl- styrene toluene	CS480442 CB419C12SA CB420SA CB48 C788	×	16/21.5 104/2.2		3.		1. 0.082 0.0002	14.3	
99990 121142 606262 5103719 121448 1330207	teluene, p-mitre- teluene, 2,4-dimitro- teluene, 2,6-dimitro- trans-chlerdene triethylamine	C7N7402 C7N6A204 C7N6A204 C10N6C16 C6A15N	***	14/12.	14/21.	5 6.				5
92476	e-rylene	CONTO		104/1.4 c/0.6 14/5.7 104/0.57	2.: V a					

*Buration is shown as number of days (d) or as chronic (c).

. priority pollutant	attended and added
 Environmental Protection Agency, priority pollutan EPA health advisory 	a Hattons Academy of Cotoness - ests as advanced managed
EPA PP EPA advis	MAS seer]

⁻ meriumai Academy of Sciences, safe no adverse response level

- Matienal Academy of Sciences, acceptable daily intake

- Morial Mealth Organization standards

- Mussian standards

- Mussian standards

- water quality standards for texicity (EPA)

- carcinogen

- hazzrious compound not designated as a priority pollstant

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APPENDIX B

LETHAL DOSES USED IN THE CALCULATION OF SCREENING CONCENTRATIONS FOR TOXICITY

The screening concentrations for toxicity are calculated as the product of an oral LD50 for a mammalian species (median lethal dose to 50% of a population of laboratory animals), a risk factor for establishing a threshold concentration for systemic toxicity in humans, body weight of the average soldier, and drinking-water consumption (see section entitled "Screening Concentrations for Toxic and Organoleptic Responses" for a discussion of this procedure). The LD50 data in Table B-1 were obtained from a computer tape of the Registry of Toxic Effects of Chemical Substances (RTECS). For each substance we selected the lowest oral LD50 for a mammalian species. If an oral LD50 was unavailable. we used an LD50 resulting from an intraperitoneal administration. If neither oral nor intraperitoneal doses were available, we used the lethal dose low (LDLo), which is defined as "the lowest dose (other than LD50) of a substance introduced by any route, other than inhalation, over any given period of time in one or more divided portions and reported to have caused death in humans or animals." We restricted our use of the LDLo to those resulting from oral doses to animals. Table B-1 also contains the cited reference in the CODEN abbreviation form reported in the RTECS. All of the CODEN abbreviations are found in Vol. 2 of the 1980 RTECS report.² In Table B-2 we have defined the various acronyms for species and exposure mode used in Table B-1.

!	i i
ORIGINAL REFERENCE**	
23	SERVICE SERVIC
HODE	######################################
LETHAL DOSE#	4808
N	ACETIC ACID, METHYL ESTER ACETIC ACID, METHYL ESTER ACETONITRILE ANILINE ACETONITRI ACETONITRI ANILINE
910	L L L L L L L

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DISC2-CHLORGETHYL
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DISCROPYL S BUTADIENE, MEXACHLORGS BUTAN-2-ONE
S BUTAN-2-ONE
S BUTAN-2-ONE
S CARBON DISULFIDE
TORESOL, 4,6-DINITROTORESOL, M--CHLOROS CRESOL, M--CHLOROS CRESOL, M--CHLOROS CURENE
S CURENE
S CURENE
S CULOHEXANE
TO CYCLOHEXANE
TO CYCLOHEXANE
TO DISULTY TIN DILAURATE
TO DISULTY TIN DILAURATE
TO DISULTY AMINE
S DIETHYLAMINE
S DIETHYLAMINE
S DIMETHYLAMINE
S DIMETHYLAMINE ENYLANINE, W-HITROSO-SUBSTANCE CASID

Table B-1. (Continued).

55.63 63.59 63.59 701-E-C-72-32 . 3- DICHLORG-- BENZGGUINONE DIOXIME - DIOXANE - PHENYLENEDIAMINE Table B-1. (Continued). CASID

Table B-1. (Continued).

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RIGINAL REFERENCE**	$\begin{array}{cccccccccccccccccccccccccccccccccccc$
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SPECIES	. SARTHREE STREET STRE
HODE	######################################
LETHAL	
SUBSTANCE	PCCB (ARGCLOR 1254) PCCB (ARGCLOR 1221) PCCB (ARGCLOR 1221) PCCB (ARGCLOR 1222) PCCB (ARGCLOR 12222)
i U	$\begin{array}{cccccccccccccccccccccccccccccccccccc$

Table B-1. (Continued).

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AL MODE SPECIES ORIGINAL E+ REFERENCE++	77E+02 ORL RAT MTIS++ PB214-270 30E+02 ORL RAT MTIS++ PB214-270 60E+02 ORL RAT AMINBC 4119,51 61E+02 ORL RAT AMINBC 4119,51 61E+02 ORL DOM AJPHAF 153(1),41,48 62E+03 IPR RAT COREAF 243,609,56 50E+03 ORL RAT AMINAB 14,587,56	
REFE	CATTORNA TO THE CATTORNA THE CA	! ! !
MODE SPECIES	RRDERAGE PATENTA PATEN	! !
MODE	20000000 20000000000000000000000000000	
LETHAL	-w4-8884 -w	
CASID SUBSTANCE	INITRO- RO- ILOTRIACETATE	
CASID	8 1 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	• 1.DIA

** Original reference CODEN from Registry of Toxic Effects of Chemical Substances (RTECS).²

Note: For each substance we selected the lowest LD50 for a mammalian species. If an cral LD50 was unavailable, we used an LD50 resulting from an intraperitoneal administration. If neither oral nor intraperitoneal doses were reported, we used the lethal dose low (LDLo).

Table B-2. Abbreviations used in Table B-1 for exposure mode and species.

Abbreviations	Definition	
ORL	Oral administration	
IPR .	Intraperitoneal administration	
RBT	Rabbit	
MUS	Mouse	
GPG	Guinea pig	
DOM	Domestic animal	
CTL	Cattle	
MAM	Mammalian species	

FOR APPENDIX B

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APPENDIX C

TASTE AND ODOR THRESHOLDS

Table C-1 contains the taste- and odor-threshold data that we acquired on organic substances that are potential contaminants in field-water supplies. The data were gathered from a number of different sources, and consequently the threshold values between substances are not strictly comparable because of different methods for determining thresholds. Another artifact of the taste/odor studies is the statistic used for reporting the organoleptic data. In some studies a minimum value is reported; in others an average value. Under the column entitled "TYPE" we have designated whether the concentrations were a minimum (M) or average (A); taste (T) or odor (O) value, based on our review of the reference. Some threshold concentrations in water were calculated from odor thresholds in air using Henry's law constant; these calculated values are indentified by a "C" in the column entitled "TYPE."

Table C-1. Reported taste- and odor-detection threshold concentrations (mg/L) for organic chemicals for which we compiled water-quality monitoring data.

CASID	SUBSTANCE	TASTE/ODOR THRESHOLD	TYPE	REFERENCE ^{F*}
67641	SUBSTANCE ACETORE ACETORE ACETORE ACETORE ACETORE ACETORE ACRYLAMIDE ACRYLONITRILE ANILINE BENZALDEHYDE BENZENE, 1,2-4-TRICHLORO- BENZENE, 1,2-DICHLORO- BENZENE, 1,4-DICHLORO- BENZENE, CHLORO- BENZENE, CHLORO- BENZENE, HEXACHLORO- BENZENE, NITRO- BIPHENYL BUTADIENE, HEXACHLORO- CRESOL-M; 6-CHLORO- CRESOL-M; 6-CHLORO-	5.00E+00 2.00E+01 4.00E+01 4.09E+01 2.40E+02	OM OAC OM OM	VANG77 AMO083 VANG77 VANG77 VANG77
98862	ACETOPHENONE	2.65E+02 6.50E-02	OM OM	VANG77 VANG77
107028	ACROLEIN	1.70E-01 1.10E-05 1.10E-01	ÖM ÖM ÖAC	ÝÄNG77 FAZZ78 AMOG83
79061 107131	ACRYLAMIDE ACRYLAMITEILE	4.005+0; 3.105-01 9.105+00 1.865+01	TM CAC CAC OM	FAZZ78 AMOO83 AMOO83 FAZZ78
62533	ANILINE	1.86E+01 6.50E+01 7.01E+01	OM OAC OM	VANG77 AMOO83 VANG77
100527 71432	BENZAL DEHYDE BENZENE	3.00E+03 3.50E+01 2.00E+01	OM OM OAC OM OM	VANG77 VANG77 AMOO83 VANG77 VANG77
120821 95501	BENZENE, 1,2,4-TRICHLORG- BENZENE, 1,2-DICHLORG-	3.135+01 5.00E-03 1.00E-02	OM OM OM	VANG77 FAZZ78 VANG77 VANG77
106467	BENZENE, 1,4-DICHLORG-	Z.40E-02 3.00E-02 3.40E-02	OAC OM OM	AMOUS3 VANG77 VANG77
108907	BENZENE, CHLORG-	1.10E+01 5.20E-02 1.00E-01	ÖÄC DAC OM	AMODES AMODES VANG77
100414	BENZENE, ETHYL-	1.00E-01 2.90E-02 1.00E-01 1.40E-01 2.00E-01	OM OAC TM M M	VANG77 AMOD83 FAZG73 VANG77 VANG77 VANG77
118741	BENZENE, HEXACHLORG-	1.40E+02 3.00E+00	OM OM	FAZZ78 VANG77
98953	BENZENE, NITRO-	3.00E+00 2.00E-01	OM OM	VANG77
92524	BIPHENYL	1.10E+01 2.00E-04	DAC DM	AMOO83 VANG77
87683 75150	BUTADIENE, HEXACHLORG- Carbon Disulfide	5.00E-04 6.00E-04	DAC DAC DAC	AMOD83 VANG77 AMDD83
59507	CRESOL-M, 6-CHLORO- CRESOL-M, P-CHLORO-	2.00E-02 3.00E+00	OM OM	VERSOS EPABOS EPABOS

Table C-1. (Continued)

CASID	CRESCL-G, 4-CHLORG- CUMENE CYCLGHEXANE CYCLGHEXANGL CYCLGPENTADIENE, HEXACHLGRG- DIETHYLAMINE DIMETHYLAMINE DIMETHYLAMINE DIMETHYLAMINE DIMETHYLAMINE DIMETHYLAMINE DIMETHYLAMINE DIMETHYLAMINE DIMETHYLAMINE DIMETHYLAMINE DICHLGRG- ETHANG, CALGRG- ETHANG, CALGRG- ETHANG, CALGRG- ETHANG, CALGRG- ETHENE, TETRACHLGRG- ETHENE, TETRACHLGRG- ETHENE, BIS(2-CHLGRGTETHYL) ETHER, BIS(2-CHLGRGTSOPROPYL) FORMAL DEHYDE FURAN METHANE, TETRACHLGRG- METHANE, TETRACHLGRG- METHANE, TRIBROMG- METHANE, TRIBROMG- METHANE, TRIBROMG- METHANE, TRIBROMG- METHANE, TRIBROMG- METHANE, TRIBROMG- METHYLAMINE NAPHTH-2-GL NAPHTHALENE	TASTE/ODOR THRESHOLD	TYPE	REFERENCE**
98828	CRESOL-O, 4-CHLORO- CUMENE	1.80E+00 8.00E-04	GM GAC	EPASOS AMODS3
110827	CYCLGHEXANE	1:00E-01 1:10E-02	OM BAC	VAÑG?? AMGG83
108930	CYCLOHEXANOL	2.00E+02 4.00E-01	OM OM	VANG77 VANG77
		2.80E+00	ŽÃC	AMO SAS
77474	CYCLGPENTADIENE, HEXACHLGRO-	1.006-03	ŎM	(ILLI)
102427	DIETHYLAMINELLIE	4:206+01	ŽÃŽ	AMOOSS
124403	DIMETHYLAMINE	2.906+01	QAC	AM0083
107062	ETHANE, 1,1,2,2-TETRACHLORG- ETHANE, 1,2-DICHLORG-	5.00E-01 2.90E+01	SM SM	FAZZ78 VANG77
75003 67721	ETHANE, CHLORO- ETHANE, HEXACHLORG-	1.905+01	ÖÄC	AMOOS 3
		1:005-02	ğM TAC	FAZZ7
64175	ETHANOL	1.005+02	<u>aw</u>	VANG27
		7:405+02	ğM _	VANG77 VANG77
		7.60E+02 8.00E+02	DAC OM	AMODA3 Vang77
75354 540590	ETHENE, 1,1-DICHLORO- ETHENE, 1,2- DICHLORO-	1.50E+00 2.60E+01	ĐÁC ĐÁC	AMOOS 3
75014	ETHENE, CHLORG- (VINYL CHLORIDE)	3.40E+00	ÖÄČ	AMOOS 3
7001/	ETHENE TRACHIABA	3:00E-01	<u>om</u>	VANG77
111444	ETHER, BIS(2-CHLORGETHYL)	7.30E-01	ÖM _	VANG77
		3.60E+01 3.60E+02	DAC Om	AMOO83 FAZZ78
108601	ETHER, BIS(2-CHLORGISOPROPYL)	2.00E-01 3.00E-01	OM OM	VANG77
50000	FARMAL DENYDE	3.20E-01	ÖM	VANG77
110009	FURAN HERTOL AL COMOL	4.00E+00	ŎM	VANG77
75310	ISOPROPYLAMINE	9.00E-01	DAC	AMOO83
75092	METHANE, DICHLORD-	1.00E-01	QAÇ	AMOD&3
56235 75252	METHANE, TETRACHLORD- METHANE, TRIBROMO-	5.20E-01 3.00E-01	OAC Om	AMOO83 Vang77
67663	METHANE, TRICHLORG-	1.00E-01	GA GAC	ZOETSO
74205	METHY! AMTNE	1.20E+01	ŤM	ALEX82
1 3 5 1 2 3	NAPHTH-2-01	1.295+00	<u>am</u>	VANG77
71203	MACHINALENE	3.00E-03 3.00E-01	OM	VANG77 VANG77
		6.80E+00 6.80E+00	OM OM	VANG77 FAZZ78
X5018	PHENANTKRENE	2.10E+01 1.00E+00	ÖÁC GM	AMOOSS VANCEZ
			V .,	1010//

Table C-1. (Continued)

CASID SUBSTANCE	TASTE/ODOR THRESHOLD	TYPE	REFERENCE**
SUBSTANCE 108952 PHENGL 58902 PHENGL, 2,3,4,6-TETRACHLORG- 933755 PHENGL, 2,3,6-TRICHLORG- 95954 PHENGL, 2,4,5-TRICHLORG- 88062 PHENGL, 2,4,6-TRICHLORG- 120832 PHENGL, 2,4-DICHLORG- 87650 PHENGL, 2,6-DICHLORG- 95578 PHENGL, 2,6-DICHLORG- 95578 PHENGL, 2-CHLORG- 106489 PHENGL, 3,4-DICHLORG- 106489 PHENGL, 4-CHLORG- 108430 PHENGL, 4-CHLORG- 108430 PHENGL, M-CHLORG- 178865 PHENGL, M-CHLORG- 178875 PROPANE, 1,2-DICHLORG- 178875 PROPANE, 1,2-DICHLORG- 108883 TOLUENE	3.00E-01 1.00E+000 4.20E+000 5.20E+000	TM GM GM GM GM GAC	EPASOS VANG77 VANG77 VANG77 VANG77 FANZ78 AMOSS
58902 PHENOL, 2,3,4,6-TETRACHLORG-	1:00E-03	18	AMOTAS FAZZ7R EPASOK
933755 PHENOL, 2,3,4-TRICHLORG- 95954 PHENOL, 2,4,5-TRICHLORG-	3.00E-01 1.00E-03	ëm Im	EPASUK EPASUK
88062 PHENOL, 2,4,6-TRICHLORG-	2.00E-03 2.00E-03 1.00E-01	YM TM OM OM	KKKKKJKJKJ UNODOBOBOBO GBGBBBBBBBBBBBBBBBBBBBBBBBBBBB
120832 PHENOL, 2,4-DICHLORO-	1.00E+00 3.00E+04 2.00E-03	TM TM OM OM	VERSSS EPASOL VANG77 VANG77
583788 PHENOL, 2,5-DICHLORG-	2.00E-04	TM	FAZZ78 EPASOK
87650 PHENGL, 2,6-DICHLGRG-	3.30E-03 2.00E-04	OM IM	EPASOK EPASOK
95578 PHENGL, 2-CHLGRG-	3.30E-03 3.30E-04 9.00E-01	OM OM TM	EPA80K EPA80K FAZZ78 FAZZ78
88755 PHENOL, 2-NITRO- 95772 PHENOL, 3,4-DICHLORG-	1.00E+01 3.00E-04	OM OM IM	FAZZ78 VANG77 EPA80K
106489 PHENGL, 4-CHLORG-	1:005-04	TM	EPASOK EPASOK
108430 PHENOL, M-CHLORD- 87865 PHENOL, PENTACHLORD-	3.30E-02 1.00E-04 3.00E-02	OM OM TM	EPA80K EPA80G EPA80G
78875 PROPANE, 1,2-DICHLORG- 113861 PYRIDINE 108883 TOLUENE	9.30E-03 9.50E-01 4.20E-02 1.20E-01	OM GAC GAC TM	EPASOQ VANG77 AMDO83 AMDO83 ALEX82
121448 TRIETHYLAMINE	1.00E+00 4.20E-01	OM UAC	ALEX82 VANG77 AMOU83

^{*} O = odor, T = taste, M = minimum concentration detectable by individuals with sensitive sensory ability, A - average concentration detectable by individuals with average sensory

ability, and C = calculated from odor thresholds in air using Henry's law constant.

** References appear as CODEN abbreviations; the CODEN abbreviations correspond to those listed alphabetically in the reference section for Appendix C.

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APPENDIX D

DATA ON THE CONCENTRATIONS OF ORGANIC SOLUTES IN FRESH AND SALT WATERS

The data on the concentrations of organic solutes in natural waters are presented in two tables: Table D-1 contains concentration data for fresh waters, including surface and ground waters, and Table D-2 contains data for salt waters. Each table contains the type of water measured; the statistical value reported; the number of samples, if available; the reference identifier; and the geographic location. The acronyms for the water types are as follows: DW = drinking water, FDW = finished drinking water, SW = surface water, GW = ground water, and SALT = salt water. The complete literature citation for each of the reference identifiers is located in the reference section of this appendix. Two types of quality control were used to ensure that the values reported in the computerized data base were correct. First, we compared the values in the data base with those in the original reference to see if they agreed; second, we statistically analyzed the values in the data base to identify values that were very high or very low. Once these outliers were found, we reviewed the source articles to determine whether the reported concentrations were valid.

Table D-1. Fresh-water concentr

	TYPE	. (MEDIAN	MAXIMUM	20	REFERENCE	PLACE	i
		CON	NTRATION	MG/L	n			
ACENAPHTHENE	TTELL TOTAL TAXENT	i <u>.</u>		0 0 0 0 0 0 0 0 0 0	Ţ	MENONA ME	CANADA CANADA CANADA CANADA	į
	ririr Saga XXXX	0.000 0.000 0.000			- (IND DA, E ONTA	
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ACENAPHTHYLENE		0000 MMMM					SA. M CATA	
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ANILINE, 2,3-DICHLORG-	XXXX ก็ตั้งก็ต่	. 20K-0 . 97K-0 . 00K-0	. 50E-0 . 70E-0	22.00 0000 0000			ARA ARA ARA ARA ARA ARA ARA ARA ARA ARA	
ANILINE, 2,4-DICHLORG-	XXXX	0000 0000 0000 0000 0000		0000			ANA	
ANILINE, 3,4-DICHLORO-	N. N. N. N.	0000 0000 0000	50E-0 90E-0	24444			METHERLANDS METHERLANDS METHERLANDS METHERLANDS	
ANILINE, 3,5-DICHLORG-	NNNV		0.9	200 E			RLAND RAND RAND RAND RAND RAND RAND RAND R	
ANILINE, 4-METHYL-	, NOW		t d	. 500m			REAND	
ANILINE, M-CHLORD-	ryki Kyri		5.00E-04 1.20E-05	.500 -000 -000 -000			RAND AND	
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Table D-1. (Continued)

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SM REFERENCE CONCENTRATIONS MG/L 9.00E-04 5.80E-04 5.00E-03 7.00E-03 7.00E-03 1.10E-03 7.40E-04 MMG-000 9.50E-04 5.80 1.70 2.70 .40E-04 1.90E-03 1.70E-03 2.70E-03 2.20E-03 2.70E-03 6.00E-04 6.60E-04 9.40E-04 7.40E-04 9.70E-04 8.70E-04 AVERAGE --ACCENTANTAL CONTROL OF BENZENE, 1,4-DICHLORG-BENZENE, CHLORONITRO-HEXACHLORG-NITRO-TOLUENE, O-CHLORG-BENZENE, CHLORO-SUBSTANCE BENZENE, ETHYL-BENZENE, BROMG-BENZENE, BENZENE,

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Table D-1. (Continued)

SUBSTANCE	TYPE	į i	MEDIA	IX	80.	REFERENCE	PLACE
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BENZENE, PROPYL- BENZGAMÍNGTRIFLUGRIDE	E S S	7.00E-0 3.00E-0	9.80E-04	9.80E-04	321		UNITED STATES
BENZGIC ACID, O-HYDROXY-	XX.					A.A.	AN
DENZGIC ACID, P-HYDRGXY-	XXX	.90E-0				~~~	LAPAK SPAK SPAK SPAK SPAK SPAK SPAK SPAK
BENZOIC ACID, M-HYDROXY-	i v	-006 -006					EXEC SAC SAC SAC SAC SAC SAC SAC SAC SAC SA
BENZG [A JANTHRACENE		. 000E				~	CANADA
	inn,					BENG79 FURU83 FURU83	LANIS LANIs LANIS LANIs LANIS LANIS LANIS LANIS LANIS LANIS LANIS LANIS LANIS LANIS LA LANIS LANIS LANIS LANIS LANIS LANIS LANIS LANIS LANIS LANIS LAN
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		. 60E-0			•	.000	E ONTARI
		000 mmr					CANADA, E ONIARIO CANADA, E ONIARIO CANADA, E ONIARIO
BENZO[A]FLUORENE	Z.	206-0			***		E ONTARI
BENZGCAJPYRENE	Z Z Z	000 mmr			ه سه سه ه	JAJA	FINIAND JAPAN
	N. N.	000 1000 1000				ana.	LAPAN Lapan Karak
	, K.	00%				مرمرم م	. THAMES
BENZOLBIFLUORANTHENE	A T					10100	
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SUBSTANCE	TYPE	AVERAGE	MEDIAN	MAXIMUM	200	REFEREN	ICE PL	ACE
		20	NTRATIONS				•)
	Y X X Y	10-04 10-04 10-06				1222	APAN NGLAN NGLAN	THAMES R
BENZOLJ JFL UGRANTHENE	TA:	040 000 000 000				788	ENHARD AND AND	MAMES
BENZO (K)FLUORANTHENE	Z X X X				-	多りと	SCENE SCENE	THAMES R
		0000				2XXX 2XXX 3XX 3XX 3XX 3XX 3XX 3XX 3XX 3X	PERCHANG PINCARK NICKAR NICKAR NICKAR NICKAR NICKAR NICKAR NICKAR NICKAR NICKAR NICKAR NICKAR NICKAR NICKAR NICKAR NICKAR NICKAR NICKAR NICKARK NICKAR NICKAR NICKAR NICKAR NICKAR NICKAR NICKAR NICKAR NICKARK NICKAR	HAMES
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BIPHENYL		000				222	CANADA CANADA CANADA	
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		000				:22	NADA NADA	ONTARI ONTARI
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BISPHENGL A	in.	. 90E-0				アアア	NA PARA	ONTARI
CHRYSENE	K X X	0.300		3.90E-03 2.50E-05	98 17	× 00	CANADA.	ONTARIO TATES, MY
	NN N				t t (222	PAN PAN	•
		. 70E-0				2000	JAPAN Denmark Finland	

Table D-1. (Continued)

SUBSTANCE	TYPE	- 1	MEDIA	XIMUM	2	REFERENCE	7	ACE
		CONC	ENTRATIONS	HG/L	SAMP			
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CRESOLS	E E E				-	ATS77 CH081	Japan Normay	
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DIBENZOTHIOPHENE		200 100 100 100 100				VES82	FINL AND NORMAY	
DIBENZG(A, H)ANTHRACENE	Z X					VES82 VES82 URU83	DENMARK FINLAND JAPAN	
ETHANE, 1,1,1-TRICHLORG-	ZZ NVI	.90E-0		.20E+0	-0-	OGUR75 MAGG77 CORR77	JAPAN ENGLAND NETHER	6
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	r S	6.90E-03			-	01887 01883	a ini	
	MAN.		1.20E-03	.20E-0		EST84 EST84	α	ATE ATE
	I A		. 00E-0	3.10E-02 1.80E-03	098 1886 1886	EST84	IAIA	TATES
	553	.10E-0		0-300		GET80 PAS1	# 10	DS ATES
ETHANE, 1,1,2-TRICHLORO-	K T	7.10E-63		.70E-0	17	AISSS	10011	(*)
			20E	20E	SIN	EST34 EST34	1010	ATE
ETHANE 1.2-DICHIORG-			104.	20E-0	1380 1360	EST34		TATES
			. 80E-0	2.00 1.00 1.00 1.00	NG	ESTS4		ATE
ETHANE, TRICHLOROTRIFLUORG- ETHENE, 1,1-DICHLORG-	.33	1.35E-01 9.62E-02		. 36F-0	10	FST84 PA81		ATE ATE
ETHENE, 1,2- DICHLORG-	y S	.32E-0		3.50E-03	98	MC 2000	A - 1	DUTARIO
	MACH WAS WAS		1.30E-03 2.70E-03	1.20E-02	321	100 100 100 100 100 100 100 100 100 100	AIAI	ET TATES TATES

₹ ₹ STATES STATES STATES N. ONTARIO LANDS STATES, N CUNITIEED STATE OF ST STATES REFERENCE NETATANA SANGRANG SAN 132 186 132 186 9868 1988 321 280 186 17 CONCENTRATIONS MG/L
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1.10E-03 2.00E-03 -04 -03 2.200 -64646 000040 こるこち?! 2.70E-03 1.10E-03 minim 9000 4444 2000 **NOME** 3.60E-05 3.75E-01 AVERAGE 7 CHLORG- (VINYLCHLORID) **TETRACHLORD** TRIBROMO SUBSTANCE ETHENE, ETHENE, ETHENE, ETHENE,

SCOTIA DDENNAMENT CONTRIBUTED STATE OF STATE O PLACE REFERENCE 22.02 20.03 20.03 20.03 .10E-05 9.70E-05 CONCENTRATIONS MG/L **EMOROW** N-4rem ---E-02 AVERAGE 600 NAVANAVALETENATURANAVANAVA ZZZZZZZZZZZZZZZ ETHER, BIS(2-CHLORGISGPROPYL) DIPHENYL TETRACHLORGISOPROPYL Table D-1. (Continued) ETHENE, TRICHLORG SUBSTANCE FERULIC ACID FL UORANTHENE

SUBSTANCE	TYPE	AVERAGE	MEDIAN	MAXIMIM	16		
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			ENIKALI	MG/L			
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METHANE, BRUNDCHLORG- METHANE, BRONDICHLORG-	ZZ.	, 08E-0		1.00E-05	96	AIS8	ANADA,
	K X 3					ISES SES	SRAEL
	, ran	1.50E-03			•	minin Minin	SAS
	NO.	766-0			,- -,		RAZ
	N.Y.	15E-0					RAZ
	3	. 45E-0			-	SES SES	ZE ZE

SUBSTANCE	TYPE	AVERAGE	MEDIA	MAXIMUM	S.	REFERENC	E PLACE
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				00. 000 113 00	80 <i>/</i>	ATAMUMUMUMUMUMUMUMUMUMUMUMUMUMUMUMUMUMUM	OFZ

SUBSTANCE	TYPE	AVERAGE	MEDI	MAXIMUM	NO.	REFERENCE	PLA	ICE
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	NNNNN LEEEE	00000			6- 6- 6- 6-		(V) 2	14
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HETHANE, TRIBROMG-	in Karaba	2.30 2.30 3.40 4.00 4.00 7.10 7.10 7.10 7.10 7.10 7.10 7.10 7		. 50 . 50 . 50 . 50 . 50 . 50 . 50 . 50	98 11		- ALAIAI	ZOVEZ ZOVEZ

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PLACE		NEW ACAPTICA ATTECA ATT
REFERENCE	,	LTTTTTTTTTTTTTTTTTTTTTTTTTTTTTTTTTTTTT
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MAXIMUM	MG/L	
MEDI	ENTRATI	พเซน
AVERAGE	CO	
TYPE	, 8 8 8 9 9	NAME OF THE PROPERTY OF THE PR
SUBSTANCE		METHANE, TRICHLORD-

SOUCHTHE SOU PLACE REFERENCE BEBERTALY AND THE STATE OF STA AVERAGE MEDIAN MAXIMUM EDIAN EDIAN MAXIMUM EDIAN EDIAN MAXIMUM EDIAN EDI .10E-04 .10E-03 .70E-05 9W49GWGVW4A **トキーろらのらしゅうご** mmm 0000 www 4.20E-06 2.60 9099999 9009994 AVERAGE MOS/SOM METHANE, TRICHLORGFLUGRG-N. N. DIMETHYLANILINE SUBSTANCE NAPHTHAL ENE MUSK XYLENE MUSK KETONE

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SUBSTANCE	TYPE	AVERAGE	MEDIA	MAXIMUM	2. 2. 2.	REFERENC	E	ACE
		CONCE	NTRATIO	2	Ę			
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PETROLEUM HYDROCARBONS PHENANTHRENE, ETHYLMETHYLENE-	NN NE				9	XXXIIII XXXXIII XXXXII XXXXIII	ENGLAND, CANADA, DENMARK	THAMES R THAMES R ONTARIO
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PHENANTHRENE, 1-METHYL-		24-00-00-0-0-0-0-0-0-0-0-0-0-0-0-0-0-0-0			Ann han fan den den den	XXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXX	FIGURE STATES OF THE STATES OF	,
		4w-ov-4			~	MVENO BEENO79 BEENO79 BEENO79 BEENO79	CCANADA CCANADA CCANADA CCANADA CCANADA	E CONTARIO CONTARIO CONTARIO CONTARIO NA PRIO
PHENANTHRENE, 2-METHYL-	rri XXXX	440 400 000 000 000 000			4	BENOT9 KVES82 KVES82 KVES82	CANADA, DENMARK DENMARK FINLAND	E ONTARI
PHENANTHRENE, 3-METHYL-		80000 0000 0000			ter ter ter	KVES82 KVES82 KVES82 KVES82	NORMAY DENMARK DENMARK FINLAND	

Table D-1. (Continued)

SUBSTANCE	•	VERAGE	MEDI	MAXIMUM	SQ.	REFERENCE	PLA	ČE
		CONC	ENTRATIONS M	MG/L				
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PHENANTHRENE, 4-METHYL-						ESSENSION OF THE SECOND OF THE	NORMAND DENMARK DENMARK	
PHENGL	E SAN	22-29-20-20-20-20-20-20-20-20-20-20-20-20-20-			han dang dang dan	MULTURA MULTUR	FINLAND NORWAY NORWAY NORWAY	
	N N N N	0000 0000 0000		. 50E-0		888×	NORWAY NORWAY NORWAY	
PHENDL, 2,3,4,5-TETRACHLORD-PHENDL, 2,3,4,6-TETRACHLORD-	ZZ.	6.00E-05	1	14.00 19.00 19.00		GA7	THERLA	
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	KKK KKK K		6.00E-05	2000 2000 2000 2000 2000 2000 2000 200			ETHERLA ETHERLA ETHERLA	
PHENOL, 2,3,4-TRICHLORG- PHENOL, 2,3,5,6-TETRACHLORG-	N.N.N.		.00E-0			2000 2000 2000	THERLA THERLA THERLA	2000
	EXE		3.00E-05	000		GM7	ETHERLA	ロロロ
	EX3		2.00E-05	. 10E-0		GA7	ETHERLA ETHERLA	
	N.		2.00E-05				ETHERLA	
	ZZ:		3.00E-05	. 70E-0			ETHERLA ETHERLA ETHERLA	
FRENOL, 2,3,5-IRICHLORD-PHENOL, 2,3,6-TRICHLORO-	X X X			000 000 000 000 000		GA7	ETHERLA	200
, 2,3-	N.W.				•		ETHERLA ETHERLA	
FILENOL, 2,4,5-TRICHLORG-	XX.				•		ETHERLA FTHERLA	201
	Z.Z.		9.00E-05	. 10E-0			THERLA	90
	XX		1.50E-04	. 40E-0			ETHERLA ETHERLA FTHERLA	906
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	K K			. 60E-0 . 00E-0		EGM7	ETHERLA	

Table D-1. (Continued)

SUBSTANCE	TYPE	AVERAGE	MEDIA	MAXIMUM	NO.	REFERENCE	E PLACE
;		CONC	ENTRA	1/9			
PHENGL, 2,4,6-TRICHLORG-	N.S.	!		5.00F-0		ECM7	TANK TOUR
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	3		2000	. 10E-0	Z:	EGM7	ETHERLAND
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	š			0.0E-0	≰ ∄	EGM/	ETHERLAND
	S		.90E-0	50E-0	.3	S C C C C C C C C C C C C C C C C C C C	FINEKLAND FINEDIAND
	K 3		1.80E-04	. 10E-0	Z	EGM7	ETHERLAND
PHENGL, 2.4-DICHLORG-	E IX		. 00E-0	0-309.	.	EGM7	ETHERLAND
	S.			50E-0	£3	ECUL FIGURE	ETHERLAND
rnewol, Z,5-DICHLOKO-	X.			. 90E-0		EGM7	ETHERLAND
PHENOL, 2.6-DICHLORG-	r I			. 60E-0	Z:	EGM7	ETHERLAND
,	Š			20E	£3		ETHERLAND
	, a		3.006-05	. 50E-0	. . .	EGA7	ETHERLAND
	K I		1 205-04	. 50E-0	X :	EGM7	ETHERLAND
	X.		. 202.	000	£3		ETHERLAND
	X.		1.50E-04	. 50E-0	3	S C C C C C C C C C C C C C C C C C C C	FINERLAND FINERLAND
	X X		. 00E-0	30E-0	Z :	CM7	ETHERLAND
PHENDE, 2-CHEORG-	3			20000	£3	EGA 7	ETHERLAND
	Š			.00E-0	I	NA SOUTH	EIHEKLAND FTHERIAND
PHENDL, 3,4-DICHLORG-	Z Z			. 70E-0	Z:	EGM7	ETHERLAND
	X.			102-07.	EZ	EGM/	ETHERL AND
FHENDL, 3,5-DICHLORG-	X.			20E-0	I.	MY A	FINERCAND
PHENOL, 4-CHLORG-	T X			. 70E-0	Z:	EGM7	ETHERLAND
	Z.			90000	3 .3	EGM7	ETHERLAND
PERSOL MACHINES	X.			OOE-D	KZ	VECUL	ETHERLAND
	Z X	1.90E-03))	E	ATS7	APAN
	X	90E-0			£ 3	ATS7	APA
	X.		.80E-0	.90E-0		TO WAY	ATAN Findiaro
	Z 3		.90E-0	. 90E-0	3	EGM7	ETHERL AND
	Y.		005-0	200	3 3		ETHERL AND
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	XX.		. 20E-0	20E-0	Z	EGM7	ETHERLAND
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1			7.36E-04 6.30E-04	2.40E-03	3	EGE 17	ETHERLAND
rninalaie, BIS (Z-ETHYLHEXYL	EXYL) SH	1.20E-02			ĒĞ:	0TR83	NEI MEKLANDS UNITED STATES
	3	.70E-0		1 . 00 E - 01	<u> </u>	11.58 24.54	RIZONA
PHINALAIE, BUIYLBENZYL PHINALIC ACID	37	3.80E-02			JW;	PAST	MITE
STATE OF THE SAUTHO	Š	.90E-0			EX	2 L C	NAG
THISH E DI-N-BOILE	N.C	.00E-0			:0	OTR8	ALTED STATE
	;	0-30/			W)	PAB	LX.

SUBSTANCE	TYPE	AVERAGE	MEDIA	MAXIMUM	20.				:
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PHTHLATE, DIETHYL POLYCHLÖRINATED BIPHENYLS	ZY.	4.60E-03 4.20E-05		! ! ! !		EPA81	UNITED	STATES.	Z.
	N. W.	.25E-0		7.705-07	-%-	SALKO AHNOVS BURNAS	GOTALY	RIVER	•
	N.	000 1000 1000				BURN82 BURN82	FRANCE		
	, N.W.				- 6 6	BURNS2 BURNS2 BURNS2	FRANCE FRANCE FRANCE		
	333				L., 600 61	BURN82 ELZA83 El 7683	FRANCE		
	333					ELZA83 ELZA83	EGYPT		
	555					ELZA83 ELZA83 ELZA83	EGYPT EGYPT		
	N.V.	00. 00.	1.35E-04	3.80E-04		ELZA83 PUCC80 SCH081	EGYPT ITALY		
	ZZZ.					SCH031 SCH031 SCH031	NORKAY NORKAY NORKAY		
PROPANE 1.2-DICH ABA.	XX.	000		•	 ,	SCHO81	NORMAY NORMAY		
			7.00m-04 7.50m-04	1.80E-03	2.58 2.58 2.58	NENTARA MENTARA MENTARA MENTARA	ANTARCT CNITED CNITED	STATES STATES STATES	
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Table D-1. (Continued)

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PYRENE, 4-METHYL-		. 20E-0			- +	NW K	SKEDENE SKEDEN	
SYRINGIC ACID	i Si Si					ES8	FINLAND	
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TOLUIDINE, CHLORO-	N.N.	.00E-0	. 405-	Z.40E-US	158	518 1177	_=:	TATES
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	X X	. 30E-0		18E-0		200	NA S
	Z.Z.	. 20E-0		. 83E-0			ANADA
	X.	. 70E-0		.24E-0		200	ANA DAAD
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	NN	000.		. 10E-0			ANA
	EZ.			. 80E-0		700	AZA DAZA DA
	F.S.	. 10E-6		. 90E-0		700	ANAD
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	ž X	. 30E-0		. 0 . m . 0 .		700	ANAD
	X X	000		. 69E-0		700	ANA
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	N.K	. 00E-0		. 60E-0			AZ AZ AZ
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	X X	. 00E-0		. 20E-0		200	ANAD
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	ZZ Z	00.00.00		. 10E-0			ANA ANA DANA
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	K I	000		.70E-0		100	ANA
	E.	100		.01E-0		200	AZA AZA DA
	XX.	.30E-0		00E-0		700	ANAD
	XX Non	.20E-0		. 56E-0			AN A
	XX.	5.00E-03		7.100E-01		MODD/4	CANADA
UKEA Vanillic acid	N.	.90E-0		. 50E-0	22	350	SRA
	X,	.90E-0				35	APA

Table D-1. (Continued)

SUBSTANCE	TYPE	AVERAGE	MEDIAN	MAXIMUM	D. V.	REFERENCE	AVERAGE MEDIAN MAXIMUM NO. REFERENCE PLACE	
		CONC	ENTRATIONS P	はん				
XYLENE	NU NU NU NU NU NU NU NU NU NU NU NU NU N	1.906-05	4.40E-04	2.50E-03	321	MATS77 WEST84	JAPAN UNITED STATES	
	er Erss	1.00E-03	5. 90E-04	9.10E-04	200 100 100 100 100 100 100 100 100 100	AENTOS ABOUTOS ABOUTOS	UNITED STATES UNITED STATES NETHERLANDS	
	33	4.00E-01					METHERLANDS METHERLANDS	

	777	AVERAGE	MEDIA	MAXIMUM	#	REFERENCE	PLACE	
- ;		CON	ENTRATIONS	1/9	SAMP			
ARCCIOR 1254	SAL	1 4 C] { { { { { { { { { { { { { { { { { { {			CUR7	PACIFI	
	SALT	. 20E-0			-,-,	CURY	PACIFI	
	SALT	- 90E-				CURY	PACIFI	
	SALT	.00E-0				CCUR7	PACIFI	
	SALT	. 60E-0			+	CUR7	PACIFI	
	SALT	. 20E-0				CURY	PACIFI	
	SALT	. 50E-0				て	PACIFI	
	SALT	. 40E-0			- 	CUR7	PACIFI	
BENZAL DEHYDE	SALT			1.03E-05	-	AEST AEST AEST AEST	PACIFI	
	SALI	5.13E-06				AUES		JU
	SALT			. 54F-0		AUES	F.	\mathbf{c}
BENZENE	SALT	200		1.03E-05	**	ACE8		
	SALT	. 53E-0				AUE7		
	SALT	.01E-0			+	AUE7		JCJ
	SALT	30E-0		•		AUE/		
	SALT	5.00E-05				AUE7 AUE7		
	SALT			6.77	-	AUEZ	E L	3 <i>1</i> 3
	SALT	6.15E-06 5.13E-05		•		SAUE81	GULF OF ME	EXIC CCC
	SALT			1.795-04	- 1	AUES		
BENZENE, 1,2-DICHLORG-	SALT	2.05E-06		. 346.	√) -	AUEX ALEX		1 11
BENZENE, ETHYL-	SALT	.00E-0		1.03E-05	• •	ACES 188		111
	SALT	1.80E-06				AUE/		
	SALT	. 40E-0			6 00.	FUET FUET	E.	111
	SALT	.00E-0				EUE/		
	SALT	. 50E-0			· • •-	AUE7		111
	SALT	.10E-0		1		AUE7		"
	SALT	1.03E-06		5.33E-06	-	AUE8 AUE8		1/ 1/
	SALT) - 10 E -		0 272		AUE3	. u.	1/1
BUTADIENE, HEXACHLORG-	SALT	3.08F-04		5.13E-06	M	AUES AUES	## ##	
	SALT	1		1.545-05	-	ACES PICES	10 11 E	•

Table D-2. (Continued)

SUBSTANCE	TYPE	AVERAGE	MEDIAN	MANTALIM		100	!	į
	Ť		1 1 1 1 1 1 1	25176	CMAN	KELEKENUE	. FLACE	
	1		ENTRATIONS					
CYCLOHEXANE	SALT	90-	; 1 1 1 1 1	 	-	AUE	OF MEXIC	
CYCL GHEXANE, ETHYL-	SALT	5.13E-06		2.05E-05	•	AUE	F OF MEXIC	
	SALT			1.03E-05	• 1	AUE	F OF MEXIC	
CYCLOHEXANE, METHYL-	SALT	5.13E-06		.135-0	M-	AUE	F OF MEXIC	
	SALT			2.56E-05	- (AUES	F OF MEXIC	
DECANAL	SALT	1.03E-06		. 0.55-0.	√) !	SAUE81	GULF OF MEXICO	
	SALT	056-0		9 2 2 0	_	AUES	F OF MEXIC	
DECANE	SALT	.00E-0		. 0.56-	-	AUES AUF7	F OF MEXIC	
	SALT	1.50E-06				AUE7	F OF MEXIC	
	SALT	40E-0				AUEZ	F OF MEXIC	
	SALT	. 60E-0			- ,	AUE7	T OF MEXIC	
	- Tac	.30E-0			_	AUE7	F OF MEXIC	
	T TAC	0.55.0		2.05E-06	•	AUES	F OF MEXIC	
DODECANE	SALT	.008-0				AUES Alif7	F OF MEXIC	
	SALT	1.20E-06				AUE7	F OF MEXIC	
	SALT	000			, ,	AUEZ	F OF MEXIC	
	SALT	.00E-0				AUE/	F OF MEXIC	
	SALT	.80E-0			-,-	AUE7	TOTAL TOTAL	
	SALI	0.25		2.05E-06		AUES	F OF MEXIC	
	SALT	5.13F-06			 +-	AUE8	F OF MEXIC	
ETUANE	SALT			5.13E-06	-	ACES ACES	T OF MEXIC	
ETHANE, 1,1,1-TRICHLORG-	SALT	3.00E-07		705.7		AUE8	IBBEAN SEA	
-	SALT	.13E-0		. 206.	-	EAK/	TES RIVER	
ETKENE	SALT	100		2.15E-04	• ,	PUE8	. W	
	SALT	40F-0				AUES	BBEAN SEA	
	SALT	4.70E-06				AUE8	IBBEAN SE	
ETHENE, TETRACHIORG-	AALT	20000				AUEB	IBBEAN SE	
	SALT	-407.		2.60E-03	-	EAR7	HES RIVER	
FINENE TOTON DOG.	SALT				•	AUE8		
	SALT	1.03E-05		. 60E-0	, ,	ORR7	TH SEA	
	SALT			5.13E-05	•	AUE8	T OF MEX	
ETHER, DIPHENYL	SALT			.60E-0		EAR7	MES RIVER	
	SALT	. 40E-0				1001 1001 1001	ANTIC OCEA	
	SAL	. 50E-0				DDIZ	ANTIC OCEA	
	SALT	1.79E-04				DDI7	ATLANTIC OCEAN	
	SALI	.60E-0				DDI7	ANTIC OCEA	
	SALT	. 40E-0				DD17	ANTIC OCEA	
HEPTANAL	SALT	18F-0			_	SCHB	RELIC OCEA	
	SALI	136-0			_	SCH8	IFIC OCEA	
					_	SCH8	IFIC OCEA	

PACEI
PLACE REFERENCE SAMP 2.05E-04 1.00E-03 E-06 E-06 E-06 CONCENTRATIONS MG/L 2.40E-03 .05E-06 2.05E-06 5.13E-06 . 138 002 NMN 2.32.72.24 2.32.72.24 2.32.24 mmmmm 666666 vvvvv4v mmmmmm 0000000 000000 E-06 90 - UU-U AVERAGE L -049VWW X48400 089 0000000 0-000000 4 MI 03 5000 400MV4 20.000.00 さしならいがし Nin. TYPI TETRACHLORG-TRICHLORG-Table D-2. (Continued) HYDROCARBONS SUBSTANCE NAPHTHAL ENE METHANE, METHANE, LIMONENE O-XYLENE METHANE HEPTANE NONANE

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SEA PLACE REFERENCE NOS SANGER SENSO SOCIO CONTROLL SANGER SANGER SENSO SOCIO CONTROLL SANGER SENSO SOCIO CONTROLL SANGER SANGER SENSO SOCIO CONTROLL SANGER SENSO SENSO SOCIO CONTROLL SANGER SENSO ₩**~** RAGE MEDIAN MAXIMUM
CONCENTRATIONS MG/L
E-05 2.56E-05 1.13E-04 3.52E-02 E-0-7 .13E-04 3.08E-06 3.078

 V844880
 V8404880
 V840480
 V840480
 V8600
 V8600 40 54E-05 AVERAGE 1.03E-(885-136 885-136 844 900000 900000 WUUWUUUUUU-UUUUUW4W4W44W4WLUUUUU--MUMMIN **21-52-5** TYPE THE PROPOSITION OF THE PROPOSITI PETROLEUM HYDROCARBONS Table D-2. (Continued) SUBSTANCE PENTADECANE DCTANE

87

AVER Table D-2. (Continued) SUBSTANCE

PLACE

REFERENCE

SEA

0000	SUBSTANCE	TYPE		DIAN	XIMUM	# 4	REFERENCE	PLACE	
######################################		:	NO I	ENTRATIONS MG/L	i	< −			
1		24	. 40E-0				ENG7	RABIA	
1		144	. 44. 1.44. 1.44.				ENG7	RABIA RABIA	ww
		44	. 90E-0				ENG7	RABIA Rabia	ww
1.00000		4	. 20E-0				ENG/	RABIA Rabia	wu
1		A I	.10E-0				ENG7	RABIA	
AALT 22.00 CHIERA AND DAYARAN SEEA AND D		49	. 40E-0				ENG?	KABIA Rabia	بدرد
AALT 12.200		S.	.70E-0				EZG7	RABIA	w
1.888		4	.17E-0	•	5.0	6	ENGS	RABIA	ww
1.2000		Ä	. 60E-0	•	2E-0		ENGS FINDS	RABIA	w
ARABIAN SEPARATION SEP		9	.25E-0	•	16-0		ENG8	RABIA	uw
AATT 6.30E-022 AATT 7.345E-022 AATT 8.356E-023 AATT 8.356E-033 AATT 8.		3	. 10E-0		0E-0		ENG8	RABIA	w
ALT 7.50E-02 ALT 7.50E-02 ALT 6.10E-02 ALT 7.50E-02 ALT 8.66E-02 ALT 8.66E-03 ALT 7.50E-03 ALT 7.50E-03 ALT 7.50E-03 ALT 8.66E-03 AL		AP	. 50E-0)		GPG8	NDAMA	u w
ALT 6.70E-02 ALT 7.80E-02 ALT 7.80E-02 ALT 7.80E-02 ALT 7.80E-02 ALT 8.80E-02 ALT 8.80E-02 ALT 8.80E-02 ALT 9.80E-02 ALT 9.80E-02 ALT 9.80E-02 ALT 9.80E-02 ALT 9.80E-02 ALT 1.49E-02 ALT 1.49E-02 ALT 1.49E-02 ALT 1.49E-02 ALT 1.49E-03 AL		4	. 50E-0				20 20 20 20 20	NDAMA NDAMA	ww
ALT 6.70 Fig. 0.23 ALT 7. 2.20 Fig. 0.23 ALT 7. 2.20 Fig. 0.23 ALT 4.00 Fig. 0.23 ALT 5.00 Fig. 0.23 ALT 5.00 Fig. 0.23 ALT 5.00 Fig. 0.23 ALT 6.20 Fig. 0.23 ALT 7. 2.20 Fig. 0.23 ALT 8. 6. 6. 6. 6. 6. 6. 6. 6. 6. 6. 6. 6. 6.		점	. 30E-0				OPG8	NDAMA	3W I
ALT 2.00E-02 ALT 4.00E-02 ALT 5.00E-02 ALT 5.00E-02 ALT 6.00E-02 ALT 6.00E-03 ALT 7.00E-03 AL		4	. 70E-0		•		0PC8	Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z	ww
ALT 5.80E-02 ALT 4.80E-02 ALT 5.40E-02 ALT 5.40E-02 ALT 5.40E-02 ALT 5.50E-02 ALT 5.50E-02 ALT 5.50E-03 ALT 1.250E-03 ALT 1.2		1	. 00E-0					NDAMA	w
ALT 2.80E-02 ALT 4.800E-02 ALT 5.00E-02 ALT 6.800E-02 ALT 5.00E-02 ALT 6.80E-02 ALT 6.80E-03 ALT 1.20E-03 ALT 1.20E-		AP	30E-0				OP G8	NDAMA	uw:
ALT 4.30E-02 ALT 4.70E-02 ALT 6.90E-02 ALT 6.70E-02 ALT 6.70E-02 ALT 6.70E-02 ALT 6.70E-02 ALT 6.70E-02 ALT 6.70E-02 ALT 7.70E-02 AL		4					aPG8	N DAMA N DAMA	щ
ALT 2.90E-02 ALDAMAN SEA ALT 4.70E-02 ANDAMAN SEA ALT 4.50E-02 ANDAMAN SEA ALT 6.90E-02 ANDAMAN SEA ALT 6.90E-02 ANDAMAN SEA ALT 6.40E-02 ANDAMAN SEA ALT 6.40E-02 ANDAMAN SEA ALT 2.98E-02 ANDAMAN SEA ALT 2.98E-02 ANDAMAN SEA ALT 2.98E-02 ANDAMAN SEA ALT 2.98E-02 ANDAMAN SEA ALT 2.50E-02 ANDAMAN SEA ALT 2.50E-02 ANDAMAN SEA ANDAM		4	20H-0				GP G8	NDAMA	w
ALT 4.70E-02 ANDAMAN SEA ALT 6.90E-02 ANDAMAN SEA ALT 2.58E-02 ANDAMAN SEA ALT 2.58E-02 ALBOR-02 ALT 2.50C/79 MEDITERRANEAN SEA ALT 2.50C/79 MEDITERRANEAN SEA ALT 2.50C/79 MEDITERRANEAN SEA ALT 3.50C/79 MEDITERRANEAN SEA CHINA SEA ALT 3.50C/79 MEDITERRANEAN SEA CHINA SEA ALT 3.50C/79 ALT 3.50C/		7	.90E-0					AMACINA AMACIN	wu
ALT 6.90 E-02 ALT 7.90 E-03 ALT 7.		4	.70E-0				0PG8	NDAMA	սա
ALT 1000 AND AMAIN SEA AND AMAIN AMAIN SEA AND AMAIN		Ä	. 90E-0				GPG8	NDAMA	w
ALT 2.05		4	. 40E-0					A DAMA A DAMA	IJЩ
ALT 2.05 ANDAMAN SEA ANDAMAN S		74	.00E-0				OP G8	NDAMA	JW
ALT 2.586-02 ALT 2.586-02 ALT 2.586-02 ALT 2.586-02 ALT 2.586-02 ALT 1.496-02 ALT 1.496-02 ALT 1.806-03 ALT 1.806-03 ALT 1.8686-03 ALT 1.86685 ALT 1.86686 ALT 1.86697 ALT 1.86686 ALT 1.86697 ALT 1.86697 ALT 1.86686 ALT 1.86697 ALT 1.8		2	. 40E-0				orcs Orcs	NDAMA NDAMA	щμ
ALT 6.90E-03 ALT 2.49E-02 ALT 2.49E-02 ALT 1.49E-02 ALT 1.49E-02 ALT 1.49E-03 ALT 2.50C/79 MEDITERRANEAN SEA ALT 1.50E-03		7	. 58E-0			Ç	OPG8	NDAMA	JWI
ALT 2.49E-02 MEDITERRANEAN SALT 1.49E-02 MEDITERRANEAN SALT 1.49E-02 MEDITERRANEAN SALT 1.49E-02 MEDITERRANEAN SALT 1.49E-03 MEDITERRANEAN SALT 2.50E-03 MEDITERRANEAN SALT 2.50E-03 MEDITERRANEAN SALT 1.80E-03 MEDITERRANEAN SALT 1.80E-03 MEDITERRANEAN SALT 1.50E-03 MEDITERRANEAN SALT 1.50E-02 CHINA SEA CHI		7	. 90E-0			32	SOL7	L BOKAN EDITER	SEA Anfan a
ALT 1.49E-02 MEDITERRANEAN S ALT 2.50C/79 MEDITERRANEAN S ALT 2.50C-03 MEDITERRANEAN S ALT 1.80E-03 MEDITERRANEAN S ALT 8.68E-03 MEDITERRANEAN S ALT 1.50E-02 CHINA SEA ALT 3.67E-02 CHINA SEA ALT 1.6E-02 CHINA SEA		Ä	. 49E-0				2017	EDITER	ANEAN S
ALT 2.50E-03 ALT 2.50E-03 ALT 1.80E-03 ALT 8.68E-03 ALT 1.57E-02 ALT 3.67E-03 ALT 3.67E-03		4	. 49E-0				S01.7	EDIJEK FOTTED	ANEAN S
ALT 1.80E-03 BASIN ALT 8.688E-03 BEDORD BASIN ALT 1.20E-02 CHINA SEA ALT 3.67E-03 CHINA SEA ALT 1.16E-02 CHINA SEA		A!	505-0				2017	EDITER	ANEAN S
ALT 8.68E-03 CHEU79 CHINA SEA ALT 1.20E-02 CHEU79 CHINA SEA ALT 3.67E-03 CHINA SEA ALT 1.16E-02 CHINA SEA		4	80E-0				ORD/	EDGRD	ASI
ALT 1.20E-02 CHINA SE ALT 3.67E-03 CHINA SE ALT 1.16E-02 CHINA SE		4	.68E-0				HEU7	HINAS	4
ALT 1.16E-02 CHINA SE		A A L	.20E-0				HEU7	HINA	
		4	.16E-0				HEU7	N AZIT	

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CENTRALLEND MG/L.	r) .	-41	A 4	***		4	****			F F	_	•			V-	•		~		70		700	36	1-	, , , , , , , , , , , , , , , , , , , 			*		. ~ ~	,		 1	4.4	Per		·	-2	· W 4	•
	11 3.10E-	1.84E-0	11 7.46E-0	1LT 2.59E-0	1. 2.96E-0	11. 2.76E-0	11	1.T 9.74E-0	1.T 9.33E-0	31 - 4 - 6 - E - C - C - C - C - C - C - C - C - C	ALT 6.11E-0	ALT 3.77E-0	4.87E-0	ALI 1.97E-0	ALT 8.00E-0	ALT 5.13E-0	ALT 1.04E-0	4.55E-0	ALT 6.54F-0	ALT 2.95E-0	1.49E-0	3.517.0 3.517.0	1.69E-0	1.08E+0	1.24E+0	11. 3.59E-0	11 2.15E-0	1.	1.28E-0	11 4.51E-0	ALT 4.13E-0	ALT 6.66E-0	7.55E-0 31 T 2 98E-3	ALT 2.90E-0	ALT 4.45E-0	ALT 2.54E+0	ALT 2.23E+0	11 2.77E-0	ALT 2.73E-0	
	7		700	กับ	50	_เ	กังก		ι λ ί	ñ	ั้งดั	ั้ง	vi i	ñΫ	กับไ	νō (v	ñ	īvī	iva	ñű	ň	ivi	เกิด	ñ	ivā	เกิด	กับก	Ň	ñů	ั้	σi i	ñvi	ังดัง	เกิง	กับ	เกิ	า้ง	กัด	

Table D-2. (Continued)

SUBSTANCE

	APPIT TO SO THE POST	TYPE	12	MEDIAN		SAMP	REFERENCE	PLACE		•
				ENIKA	MG/L		1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1			
		בבי	.03E-0				ENG	NDIAN	CEA	•
AAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAA		الداد	. 35E-0				SEC	NA POL	SE S	
	######################################	44	.06E-0 .85E+0			.0-	ENG8	NA	SE S	
		44	. 36E+0					N A	C.F.	
		A.	. 02E+0				ENC SSC SSC SSC SSC SSC SSC SSC SSC SSC S	NOIAN	44 11 11	
		44	. 00E-0				GGE8	RCTIC	CEN	
		4	. 20E-0				OCE8	RCT IC	CEA	
A		Z,	. 00E-0				SHOOT OF THE STATE	RC1 IC	CEA	
AAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAA		ᇹ	. 90E-0				GGES	RCT IC	SE SE	
AAATT 10000	AAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAA	44	70E-0				OCE 8	RCTIC	CEA	
AAALT 10.000 EE 0.04 AAART 10.000 EE 0.000 EE 0.000 AAART 10.000 EE 0.000 AAAATT 10.000 AAAATT 10.000 EE 0.000 AAAATT 10.000 AAAA	AAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAA	A	. 40E-0					RCTIC	Since Since	
AALT 1 1 2 200	AALT 11-6-04 AALT 11-6-04 AALT 11-6-04 AALT 11-6-04 AALT 11-6-04 AALT 11-6-04 AALT 11-6-06 AART 11-6-06 AA	뒫;	.90E-0					2C-1C	ところ	
AALT 1 1 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	AALT 1000000000000000000000000000000000000	4	. 80E-0				GGE8	RCTIC	CENT TENT	
AAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAA	AALT 1000-04 AART 1000-06 AA	3-	. 10E-0				OCE8	RCTIC	CEA	
AATT 1.2006	AATT 1	A P					OCES	RCTIC	CEA	
AAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAA	AATT 1000 AAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAA	A	.30E-0		•		30 E	RCTIC	SEA	
AALT 1 1 200	AALT 122000E 004 AALT 12200E 007 AAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAA	8	, 50E-0				DCE 8	1 1 1 1 1	מנו מנו	
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AALT 1200E-024 AALT 1200E-027 AAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAA	AALT 1 1 2 2 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	14					DGES	ACTIC	CEA	
AATT 1.200E-07 AATT 10000000000000000000000000000000000	E,	. 00E-0					RCTIC	SE CO		
AALT 1.2006-07 AART 1.2006-07	ALT 1.00E-07 ARROTTIC 00CCEAN ARROTTIC 00CCE	E:	0000				0 K	11. 11. 11. 11. 11. 11. 11. 11. 11. 11.	S C	
AALT 12.000E-05 AALT 13.000E-05 AALT 13.000E-05 AALT 13.000E-05 AALT 13.000E-05 AALT 13.000E-05 AALT 13.000E-05 AART 13.000E-05 AART 13.000E-05 AART 10.000E-05 AART 1	AALT 13.20000-004-004-004-004-004-004-004-004-00	3	. 00E-0				OCES	RCT IC	יות מית	
AALT 1.200E-04 ARCTICO GCCCAAALT 2.300E-05 ARTT 1.200E-05 ARRCT11C GCCCEAAARCT 1.000E-05 ARCT 1.000E-05 ARCT 1.000E-05 ARRCT 1.000E-05 ARR	ARIT 1 200 E 20 ARROLLIO DOCUEDA ARROLLI	44	. 00E-0				OCE8	RCTIC	CEA	
AALT 1.200E-04 ARCTICO GOOGGS ARTT 1.200E-04 ARTT 1.200E-04 ARTT 1.200E-04 ARCTICO GOOGGS ALT 1.2200E-04 ARCTICO GOOGE ALT 1.2200E-04 ARCTICO GOOGE ALT 1.2200E-04 ARCTICO GOOGE ARCTICO GO	Z Z					OCEB	RC1 IC	CEA		
AALT 1.200E-04 ARCTIC GCEAA ART 7.200E-04 ART 7.200E-04 ART 7.200E-04 ART 7.200E-04 ART 1.200E-04 ARCTIC GCEAA ARCTIC GC	ALT 3.90E-04 ALT 3.90E-05 ALT 1.00E-05 ALT 1.00E-04 ARCTIC GCEAA AR	Ā	. 80E-0					RCTIC	SE SE	
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MEDIAN MAXIMUM INCENTRATIONS MG/L AVERAGE Table D-2, (Continued) SUBSTANCE

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Table D-2. (Continued)

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